



---

## Course Contents for Subjects with Code: CHEM

This document only contains details of courses having code **CHEM**.



Code	Subject Title	Cr. Hrs	Semester
CHEM-101	Chemistry-I (Physical Chemistry)	3	I
Year	Discipline		
1	Botany, Zoology, Chemistry-I, II		

### STATES OF MATTER

#### A. Gases:

Law of equipartition of energy, Collision diameter, collision number, collision frequency and mean free path; Viscosities of gases, measurements, effect of temperature and pressure on viscosities of gasses; Critical phenomenon of gases and experimental determination of  $P_c$ ,  $V_c$  and  $T_c$ ; Concept of molecular velocities of gasses according to Maxwell's distribution law and comparison of various velocities.

#### B. Liquids:

The properties of liquids like surface tension, viscosity, refractive index and dipole moment; Parachor, reheatochor and molar refraction as additive and constitutive properties; Measurement of refractive index and dipole moment; Magnetic susceptibility and its measurement by Gouys balance.

#### C. Solids:

Symmetry operations and Bravais lattices; Concept of X-Ray diffraction, Bragg's equation and crystal structure analysis; Powder method of crystal structure analysis; X-ray crystallography of sodium chloride crystal; Heat capacities of solids.

### CHEMICAL THERMODYNAMICS:

Heat capacity as  $C_p$  and  $C_v$ ; Difference in  $C_p$  and  $C_v$  and ration of  $C_p$  and  $C_v$  towards atomicity; Temperature dependence of heat capacities; Quantitative effect of temperature over enthalpy change and internal energy change; Types of thermodynamical processes; Isothermal reversible expansion of ideal gases; Adiabatic process of ideal gasses; Second law of thermodynamics, Carnot cycle, efficiency of heat engine and concept of entropy; Thermodynamics scale of temperature entropy for phase transition, spontaneity and reversibility; Entropy change in reversible and irreversible processes; Temperature dependence of entropy, Variation of entropy with pressure and volume; Concept of free energy; Derivation of Gibbs and Helmholtz equation; standard free energy of formation; Partial molar quantities, Chemical potential, variation of chemical potential with pressure and temperature fugacity; Thermodynamic of equilibrium, Reaction isohore; Calusius-Clapeyron equation; Molecular basis of entropy and probability.

### CHEMICAL KINETICS:

Derivation of kinetic expression of zero order, first order, second order (with same and different concentration) and third order reactions (with same concentrations) with suitable examples; Equation for half life periods and determination of rate constants; Methods of measurements of order of reactions giving examples with different techniques; Derivation of Arrhenius equation and measurements of Arrhenius parameters, Measurement of slopes of Arrhenius plots for some important reactions Bimolecular collision theory of reaction rates and its failures; Collision theory of uni-molecular, gas phase reactions (Lindeman mechanism); Introduction transition state theory of reaction rates.

### SOLUTION:

Thermodynamics derivation of colligates properties as lowering of vapor pressure, elevation of boiling point, depression of freezing point; Relationship between lowering of vapor pressure with  $\Delta T_b$  and  $\Delta T_f$ ; Osmotic pressure an their determination; Concept of semi



permeable membrane, Isotonic solution, theory of osmotic pressure, relationship between vapor pressure and osmotic pressure, Abnormal colligative properties describing association and disassociation of solutes; Fractional distillation and idea of azotropes in detail; Concept of colloids; Classification of Colloids; their properties with reference to dialysis, electro dialysis, sedimentation, precipitation, ultra filtrations, Suspensions and gels; Tyndal cone effect; Macromolecules and micelles.

**SURFACE CHEMISTRY:**

Introduction to adsorption; Difference between physical and chemical adsorption; Adsorption of gases by solids; Different types of adsorption isotherms with special reference to Langmuir adsorption isotherm and its applications; Freundlich adsorption isotherm giving some important applications; Brief introduction to catalysis; Theories of Catalysis; Activation energy for catalyzed reactions; Kinetics of enzyme catalysis; Theories of catalysis; Activation energy for catalyzed reactions; Kinetics of enzyme catalysis.

**Recommended Books:**

1. Adamson A. W. "Understanding Physical Chemistry" 3<sup>rd</sup> Ed., Benjamin Cummings Publishing Company Inc.
2. Akhtar M.N.& Ghulam Nabi, "Textbook of Physical Chemistry", ilmi Kutab Khana, Lahore.
3. Bhatti H.N. and K.Hussain, "Principles of Physical Chemistry"; Carwan Book House, Lahore.
4. Maron S.H. & B. Jerome, "Fundamentals of Physical Chemistry", Macruthan Publishing Co., Inc. New York. (Also published by National Book Foundation).
5. Atikins P.W.& M.J.Clugston, "Principles of Physical Chemistry" Pitman Publishing Company (1988).
6. Moore W.J. "Physical Chemistry", 5<sup>th</sup> Ed. Longmans Publishers.
7. Jones M. "Elements of Physical Chemistry" Addison-Sesky Publishing Company.
8. G.M.Barrow, International six Edition "Physical Chemistry".
9. IRA. N. Levine fourth edition "Physical Chemistry"
10. Alberty and Danials, "Physical Chemistry"
11. Castallon, "Physical Chemistry"
12. Laidler & Meiser "Physical Chemistry"
13. Friemental "Chemistry in Action"



Code	Subject Title	Cr. Hrs	Semester
CHEM-102	Chemistry Lab-I (Physical Chemistry)	1	I
Year	Discipline		
1	Botany, Zoology, Chemistry-I, II		

1. Determination of percentage composition by surface tension, viscosity and refractive index method.
2. Determination of heat of solution for solids and liquids.
3. Quantitative measurement of colored salt of  $\text{KMnO}_4$ ,  $\text{K}_2\text{Cr}_2\text{O}_7$  in colorimeter.
4. Study of first order reaction:
  - a) Study of hydrolysis of methylacetate
  - b) Measurement of rate constant
5. Measurement of molecular weight by; Depression of freezing point.
6. Determination of transition temperature of  $\text{Na}_2\text{SO}_4 \cdot 10 \text{H}_2\text{O}$ ;  $\text{Na}_2\text{CO}_3 \cdot 10 \text{H}_2\text{O}$ ;  $\text{MgSO}_4 \cdot 7 \text{H}_2\text{O}$

**Recommended Books:**

1. Crocleford H.D., H.W.Biard, F.W. Getzen & J.W. Nowell, "Laboratory Manual of Physical Chemistry", 2<sup>nd</sup> Ed., John Wiley & Sons London.
2. Das R.C. and B. Behera, "Experimental Physical Chemistry", Tata McGraw Hill Publishing Company Limited.
3. Levitt B.P., "Findlay's Practical Physical Chemistry", 9<sup>th</sup> Ed., Longman Group Limited.



Code	Subject Title	Cr. Hrs	Semester
CHEM-103	Chemistry-II (Inorganic Chemistry)	3	II
Year	Discipline		
1	Botany, Zoology, Chemistry-I, II		

### Periodicity

Modern periodic table; Similarities and differences in first row elements, their diagonal and vertical relationship with other elements; Electro negativity of elements (Pauling and Mullikan scales); Polarizability and polarizing power of ions; Periodicity in the properties of transition and inner transition elements.

### Theories of Chemical Bonding

Nature and types of chemical bonding; Modern concept of valence bond theory (VBT), molecular orbital theory (MOT) and their applications to homo and hetero di- and polyatomic inorganic molecules, explaining the conventional and modified MO diagrams; Valence shell electron pair repulsion theory (VSEPR), explaining the shapes of inorganic molecules (i.e.  $AB_2$ ,  $AB_3$ ,  $AB_2E$ ,  $AB_4$ ,  $AB_3E$ ,  $AB_2E_2$ ,  $AB_5$ ,  $AB_4E$ ,  $AB_3E_2$ ,  $AB_2E_3$ ,  $AB_6$ ,  $AB_5E$ ,  $AB_4E_2$ ) and directed valence theory (Hybridization), Metallic bonds (detailed concept).

### Acid-Base Concept

General concept of acids and bases. Detail of Lewis concept of acids and bases; Soft and hard acid-base (SHAB) concept and its applications. Relative strength of acids and bases based on  $P_k$  values. Reactions of acids and bases. Relationship between redox reactions and acid base reactions. Indicators and theory of indicators.

### Chemistry of d-Block Elements

Electronic configuration and oxidation states of transition elements. Metallurgy of chromium, nickel and copper. Theories of coordination compounds, valence bond theory (VBT), molecular orbital theory (MOT) and crystal field theory (CFT) for tetrahedral and octahedral complexes. Nomenclature and Isomerism in coordination compounds. Chelates. Application of coordination compounds.

### Nuclear Chemistry

Phenomena of radioactivity; Natural radioactivity, Radioactive disintegration series, rate of disintegration and half life period, Mass defect and binding energy, nuclear stability; measurement of nuclear radiation, Wilson cloud chamber and Geiger-Muller counter, Carbon dating; Artificial radioactivity and nuclear transformations, Nuclear reactions (fission and fusion), Uses of radioactive isotopes; Biological effect of nuclear radiation.

### Chemical Industries

Glass, Soda ash and Soap.

### Recommended Books:

1. Cotton, F, Albert, Geoffrey Wilkinson and Paul L. Gaus, "Basic Inorganic Chemistry", John, Wiley & Sons Inc, 3<sup>rd</sup> Edition (1995).
2. Lee, J.D., "Modern Inorganic Chemistry", Chapman & Hall, 5<sup>th</sup> Edition (1996).
3. Jolly, William, L., "Modern Inorganic Chemistry", McGraw Hill, 2<sup>nd</sup> Edition (1991).



4. Shriver, D.F., P.W. Atkins and C.H. Langford, "Inorganic Chemistry", Oxford, 2<sup>nd</sup> Edition (1996).
  5. Sharp, A.G. "Inorganic Chemistry", Longman, 3<sup>rd</sup> Edition (1992).
  6. Rayner Canham, Geiof., "Descriptive Inorganic Chemistry" & Co. (1995).
  7. Jefferey, G.H., j. bassett, J.Mendham and R.C. Denney, "Vogel's text book of Quantitative Chemical analysis", 5<sup>th</sup> Edition, Benjamin Cummings, (1989).
-



Code	Subject Title	Cr. Hrs	Semester
CHEM-104	Chemistry-II (Inorganic Chemistry)	3	II
Year	Discipline		
1	Botany, Zoology, Chemistry-I, II		

**PAPER CHROMATOGRAPHY**

Separation & identification of cations/basic radicals of group I, II.A, II.B & III. Also calculate their R<sub>f</sub> values.

**ARGENTOMETRY****MOHR'S METHOD**

- 1) Determine the %age purity of NaCl (rock salt)
- 2) Determine the amount of NaCl in the commercial sample of soda ash.

**VOLHARD'S METHOD**

- 1) Determination of %age purity of HCl.
- 2) Determination of silver in the given sample, using KSCN or NH<sub>4</sub>SCN.

**REDOX TITRATIONS (By using both internal and external indicators)**

- 1) Determination of amount/dm<sup>3</sup> of FeSO<sub>4</sub>.7H<sub>2</sub>O with K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>.
- 2) Determination of %age purity of K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> by using standard solution of Mohr's salt.
- 3) Determination of number of water molecules (x) in FeSO<sub>4</sub>. xH<sub>2</sub>O using K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>.
- 4) Determination of Ca<sup>2+</sup> by KMnO<sub>4</sub>.
- 5) Determination of %age of iron in ferric alum (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>.Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>.24H<sub>2</sub>O using K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>.

**COMPLEXOMETRY**

- 1) Standardization of EDTA solution by magnesium/zinc sulfate solution.
- 2) Find out the amount of Ca<sup>2+</sup> in the given sample of marble (lime stone).
- 3) Determination of Ca<sup>2+</sup> and Mg<sup>2+</sup> in the sample by using EDTA.

**Books Recommended:**

1. Vogel, "A.I.A. Text Book of Macro and Semi micro-qualitative Inorganic Analysis", Longman Green & Co., (1995).
2. Skoog, D.A., D.M. West and F.J. Holler, "Analytical Chemistry", 6<sup>th</sup> Edition, Saunders College Publications, (1994).
3. Javed Iqbal, Amin, "Theory and Practice of chromatography", Higher Education Commission, Islamabad, (2002).



Code	Subject Title	Cr. Hrs	Semester
CHEM-111	Fundamentals of Chemistry	3	I
Year	Discipline		
1	Physics		

Atomic structure, Periodic table and Atomic properties, Types of Chemical Bonding, Gaseous, and Liquid states of matter.

Nature of covalent bond, Lewis structure, bond length, bond angles and bond energies, localized and delocalized bonding resonance valence bond theory and molecular orbital concepts. Hybridizations.  $sp^3$ ,  $sp^2$ , and  $sp$  orbital. Dipole moments, inductive and resonance effects, rules for relative contribution from different resonance structures.

Modern concepts using mathematics for understanding the principles, Fundamental laws, Atomic molecular structure, states of matter. Equilibrium, Kinetic, and elementary inorganic, organic, and nuclear chemistry. Physical chemistry

**Recommended Books:**

1. Physical Chemistry by W.J. Moore, Prentice Hall College Div: 4th edition. June 1972.
2. Basic Inorganic Chemistry by Cotton Wilkinson LeGair 3<sup>rd</sup> Ed. 1995.
3. Organic Chemistry by T.R. Morrison and R N. Boyd Allyn and Bacon Inc. Longman Higher Education Division (a Pearson Education company); January 1, 1979.
4. Organic Chemistry by Khairat & Rahman. The Caravan Book House.





Code	Subject Title	Cr. Hrs	Semester
CHEM-201	Chemistry-III (Organic Chemistry)	3	III
Year	Discipline		
2	Botany, Zoology, Chemistry-I, II		

### Basic Concepts in Organic Chemistry

Hybridization of orbitals of carbon atoms in alkanes, alkenes, alkynes and arenes. Hybridization of orbitals of nitrogen, oxygen and sulfur atoms in various functional groups. Localized and delocalized chemical bonding; Conjugation and hyper conjugation; Resonance, rules of resonance, resonance energy, resonance hybrid, factor effecting the resonance; Inductive effect, applications of inductive effect and resonance on various properties of organic compounds; Steric effect and its applications, Hydrogen bonding and its effects on various properties of organic compounds, Tautomerism.

### Nomenclature of Organic Compounds

Nomenclature of alkanes, alkenes, alkynes, cycloalkanes, bicycloalkanes, spiroalkanes, Monofunctional and polyfunctional derivatives of open chain and cyclic compounds; Polysubstituted benzenes; Polycyclic hydrocarbons such as naphthalene, anthracene, phenanthrene and their derivatives; Heterocyclic compounds.

### Hydrocarbons

#### a) Alkanes and Cycloalkanes

Preparation of alkanes from alkyl halides, coupling of alkyl halide and alkylboranes, reduction of carbonyl compounds, Kolbe's electro synthesis, Corey-house-synthesis, hydrogenation of alkenes and alkynes.

Reactions of alkanes with halogens, their mechanism and comparison of reactivities of halogens; combustion, isomerization, nitration and sulfonation.

Preparations of cycloalkanes by Freund synthesis, Hydrogenation of cyclic alkenes;; Structure and stability of cycloalkanes; Reaction of cycloalkanes.

#### b) Alkenes and Alkynes

Preparation of alkenes from elimination reaction of alkyl halides and alcohols; Mechanism and orientation of eliminations; Dehalogenation of vicinal dihalides with mechanism; Pyrolytic eliminations.

Reactions of alkene; relative stability and reactivity; Addition of halogens, additions of halogen acids and the rules governing these reactions, hydration reactions, oxidation reactions including epoxidation and hydroxylation, polymerization; Simon-Smith and Diels-Alder reactions.

Preparation of alkynes by carbide process, dehydrohalogenation of dihalides and alkylation of terminal alkynes.

Reactions of alkynes: addition reactions with mechanisms, hydration reactions, oxidation, reduction, hydroboration, formation of metal acetylides, polymerization (linear and closed chain).

#### c) Aromatic Hydrocarbons

Structure of benzene, Resonance energy of benzene, Aromaticity, criteria for aromaticity, Evidences of aromaticity; Natural sources of aromatic hydrocarbons; Preparation of aromatic hydrocarbons by different methods.

Reaction of aromatic hydrocarbons: electrophilic aromatic substitution reactions i.e. nitration, halogenation, Friedel-Crafts reaction and its limitations, sulfonation; Orientation and reactivity of substituted benzenes;



Nucleophilic aromatic substitution reactions; reaction such as addition, hydrogenation, Birch reduction, and oxidation reactions of side chains.

Polycyclic aromatic hydrocarbons like naphthalene, anthracene and phenanthrene, their resonance structures and relative stabilities; Synthesis of naphthalene; Electrophilic substitution reactions of naphthalene; Oxidation and reduction reactions; Brief description of orientation and reactivity of naphthalene

### Isomerism

- **Conformational isomerism:** conformational analysis of ethane, n-butane, cyclohexane, mono- and di-substituted cyclohexanes.
- **Optical isomerism:** optical activity; chirality and optical activity; enantiomers, diastereomers; racemates and their resolution; D, L and R, S conventions; Optical Isomerism in cyclohexanes, biphenyls and allenes
- **Geometrical Isomerism:** cis and trans isomers; E-Z convention; Determination of configuration of the isomers; Inter-conversion of geometrical isomers; Geometrical isomerism in cyclic compounds.

### Alkyl halides

Preparation of alkyl halides from alcohols and carboxylic acids;

Chemical reactions: Aliphatic nucleophilic substitution reactions,  $SN_1$  and  $SN_2$  mechanism, effects of the nature of substrate, attacking nucleophile, leaving group and the nature of solvent. Elimination reactions,  $E_1$  and  $E_2$ , mechanisms, orientation of elimination (Hoffmann and Saytzeff rules).

Grignard Reagents; synthesis, structure, and reactions with active hydrogen compounds, carbonyl compounds such as aldehydes, ketones, esters, acid halides and  $CO_2$ ; Reactions with nitriles, ethylene oxide, sulphur and oxygen.

### Chemistry of Hydroxyl Group containing Compounds and Ethers

**Alcohols:** Physical properties; Preparation of alcohols by the reduction of carbonyl compound; Reactions of alcohol with metals, organic and inorganic acids; Oxidation of alcohols; Distinction between primary, secondary and tertiary alcohols; Preparation of diols, triols and their important reactions and uses.

**Phenols:** Physical properties; Synthesis of phenols; Reactions of phenols such as acylation, Friedel-Crafts reaction, nitration, sulfonation, carbonation, formylation and diazo coupling.

**Ethers:** Physical properties; Preparation of ethers from alcohols, alkyl halides and alkenes; Reactions of ethers; Brief introduction of crown ethers and polyethers.

### Recommended Books:

1. C.K. Ingold, "Structure and mechanism in organic chemistry", C.B.S.
2. I.L.Finar, "Organic Chemistry", Vol. I, Pearson Education, L.P.E.
3. I.L.Finar, "Organic Chemistry", Vol. II, 5<sup>th</sup> Edition, L.P.E.
4. Jerry March, "Advanced Organic Chemistry, Reaction, Mechanism and Structure", 5<sup>th</sup> Edition, Wiley Inter Science.
5. Morrison and Boyd, "Organic Chemistry", 6<sup>th</sup> Edition, Prentice Hall.
6. Seyhan N. Ege, "Organic Chemistry Structure and Reactivity", 3<sup>rd</sup> Edition, The University of Michigan, A.I.T.B.S. Publishers & Distributors (Regd.).
7. Thomas H. Lowry, Kathleen Schueller Richardson "Mechanism and Theory in Organic Chemistry", 3<sup>rd</sup> Edition, Harper and Row Publishers, New York.
8. Alder, Baker, Brown, "Mechanism in Organic Chemistry", Wiley Publishers.
9. Atkins Carey, "Organic Chemistry", A Brief Course, 2<sup>nd</sup> Edition.
10. Peter Sykes, "A guide book to mechanism in organic chemistry", 6<sup>th</sup> Edition, Pearson Education, Singapore.



- 
11. Carruthers, "Modern Methods of Organic Synthesis", Cambridge low Priced Edition, Cambridge.
  12. Harris, Wamser, "Fundamentals of Organic Reaction Mechanism", Wiley Publishers.
  13. G. Malcolm, Dyson, "A Manual of Organic Chemistry", Vol. I.
  14. Canant Blat, "The Chemistry of Organic Compound", 5<sup>th</sup> Edition.
  15. R. Panico, W.H.Powell, Jean-Claude Richer, "A guide to IUPAC Nomenclature of Organic Compounds", Blackwell Sci. Publication, 1993.
-



Code	Subject Title	Cr. Hrs	Semester
CHEM-202	Chemistry Lab-III (Organic Chemistry)	1	III
Year	Discipline		
2	Botany, Zoology, Chemistry-I, II		

1) **Compound Analysis**

Identification of organic compounds containing only one functional group with special emphasis on compounds containing following functional groups.

-COOH, -OH, C=O, -NH<sub>2</sub>, and -CONH<sub>2</sub>

2) **Basic Experimental techniques used in organic chemistry**

- 1) Filtration
- 2) Simple and fractional distillation
- 3) Solvent extraction
- 4) Sublimation
- 5) Re-crystallization
- 6) Column Chromatography

3) **Estimations (volumetric)**

- 1) Determination of molecular weight of a carboxylic acid.
- 2) Estimation of amide group and glucose.

**Recommended Books:**

1. K.M.Ibne Raza, M.A. Rehman, Abdur Rehman, "Organic Chemistry", The Carvan Book House, Lahore.
2. B.S. Furniss, "Vogel's T.B of Practical Organic Chemistry", Addison Wesley Longman, Inc. 1989.
3. Frederick George Mann and Saunder, "Practical Organic Chemistry", The English Language Book Society, 1960.
4. Daniel R. Palleras, "Experimental Organic Chemistry , John Willey & Sons" Inc., 2000.
5. James A. Moore, "Experimental methods in Organic Chemistry", Holt-Saunders Int., 1983.
6. R.L. Shriner, R.C Fuson, D.V. Curtin and T.C Morrill "The systematic identification of organic compounds, 6th ed. John Willey & sons, 1979.



Code	Subject Title	Cr. Hrs	Semester
CHEM-203	Chemistry-IV (General Chemistry)	3	IV
Year	Discipline		
2	Botany, Zoology, Chemistry-I, II		

#### **Quantum Mechanics and Atomic Structure:**

Elementary treatment of Compton effect and photoelectric effect; Brief discussion of result of Bohr's Model and its defects; Somerfield's modification and evolution of azimuthal quantum number; Dual nature of matter; Verification of dual nature by Davisson and Germer's experiment; Detail of Heisenberg's uncertainty principle; Postulates of quantum mechanics; Brief introduction of operators; Derivation of time independent Schrodinger wave equation in terms of polar coordinates and derivation of principle quantum number; Energy equation for free motion of particle in one-dimensional box; Eigen values and Eigen functions; normalization of wave function; Probability functions; radial distribution, probability density functions.

#### **Electrochemistry:**

Electrolytic conduction and its measurement; Specific, equivalent and molar conductance, Determination of resistance; Cell content, conductance ratio, conduction of strong and weak electrolytes; Ionic motilities and their determination; Kohlrausch's law and its applications; Faraday's law (first and second) and their significance; Transport number; Hittort's rule; Determination of transference number by Hittort's method. Applications of conductance measurement; EMF of the chemical cells; Electrode potential and its measurement with reference to Weston standard, glass electrode, calomel electrode and quinhydrone electrode; Nerst equation; Thermodynamics of cells; Concentration of cells with liquid junction and without liquid junction.

#### **Evaluation of Analytical Data and Essentials of Chemical Analysis:**

Some fundamental concepts like mole, activity and activity co-efficient; Concepts of mean, median, accuracy, precision, significant figures; Various types of errors, their detection and elimination; Standard deviation; Relative standard deviation; Confidence limits; Rounding off the analytical data. Law of mass action and its applications; precipitation and solubility product; common ion effect; Co-precipitation, fractional precipitation, concept of pH, buffers, working of indicators.

#### **Spectroscopy:**

Electromagnetic radiation and its interaction with matter; Nature of different transitions possible in atoms and molecules; Electronic, vibrational, rotational and other possible transitions by absorption of radiation by molecules and atoms. Development of spectroscopic analytical techniques employing various transitions. Classification of spectroscopic techniques on the basis of type of radiation, phenomenon occurring and the nature of the matter. Basic introduction to atomic and molecular spectroscopic techniques including flame emission, spectrophotometry, UV, IR spectroscopy.

#### **Chemistry of Carbonyl Compounds:**

Preparation of aldehydes and ketones by pyrolysis of calcium salts of acids, acylation of alkenes and arenas, reduction of acid halides and nitriles. Physical properties of aldehydes and ketones; Structure and reactivity of carbonyl group; Comparison of the reactivity of aldehydes and ketones; Nucleophilic additions of water, alcohols, ammonia and its derivatives, hydrogen cyanide, bisulfite, reduction and oxidation reactions; Aldol condensation and related reaction; Cannizzaro's reaction; Wittig reaction; Oxidation reactions, Chemical tests of aldehydes and ketones.



---

**Chemistry of Carboxylic Acids and Their Derivatives:**

Physical properties of carboxylic acids; Effects of different parameters on the acid strengths of aliphatic and aromatic carboxylic acids. Chemical properties like salt formation nucleophilic acyl substitution, reduction of carboxylic acids, decarbonylation, Hunsdicker reaction, Kochi reaction, substitution at  $\alpha$ -carbon. Preparations, properties and reactions of acids chlorides, acids anhydrides, amides, cyanides, and esters; Malonic and acetoacetic esters syntheses.

**Recommended Books:**

1. Adamson A. W. "Understanding Physical Chemistry" 3<sup>rd</sup> Ed. Benjamin Cummings publishing company Inc.
  2. Akhtar M.N. & Ghulam Nabi, "Textbook of Physical Chemistry" ilmi kutab khana, Lahore.
  3. Bhatti H.N. and K. Hussain, "Principles of Physical Chemistry"; Carwan Book House, Lahore.
  4. Shriver, D.F., P.W. Atkins and C.H. Langford, "Inorganic Chemistry"; Oxford, 2<sup>nd</sup> Ed. (1996).
  5. Snarp, A.G. "Inorganic Chemistry", Longman, 3<sup>rd</sup> Edition (1992).
  6. Rayner Canham, Gelof, "Descriptive Inorganic Chemistry" & Co. (1995).
  7. Daniel R. Paller, "Experimental Organic Chemistry, John Willey & Sons" Inc., 2009.
  8. James A. Moore, "Experimental methods in Organic Chemistry" Holt-Saunders Int. 1983.
  9. R.L. Shriner, R.C. Fuson, D.IV. Curtin and T.C. Morrill "The systematic Identification of organic compounds, 6<sup>th</sup> ed. John Willey & sons, 1979.
-



Code	Subject Title	Cr. Hrs	Semester
CHEM-204	Chemistry Lab-IV (General Chemistry)	1	IV
Year	Discipline		
2	Botany, Zoology, Chemistry-I, II		

- 1) Verification of first law of Faraday by electroplating of  $\text{CuSO}_4$ ,  $\text{NiSO}_4$ ,  $\text{Cr}_2(\text{SO}_4)_3$ ,  $\text{Ag}^+$  CN and  $\text{Au}^+$  CN.
- 2) Conductometric and potentiometric titration using conductivity bridge and pH meter.
- 3) Measurement of reduction potential using of Zn, Cu, Ag, Al etc. by using calomel electrode.
- 4) Verification of Langmuir Isotherm.
- 5) Thin layer chromatography.

**Preparations:**

- 1) Ferric Alum
- 2) Potassium tri-oxalato aluminate
- 3) Sodium Thiosulfate
- 4) Amm. Coppr (II) Sulphate

**Iodometry:**

- 1) Determination of iodide and KI in the given sample solution.
- 2) Standardization of  $\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$  solution.
- 3) Determination of amount/ $\text{dm}^3$  of  $\text{Cu}^{2+}$  in  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  using  $\text{Na}_2\text{S}_2\text{O}_3$  and KI.
- 4) Determination of number water molecules (x) in  $\text{CuSO}_4 \cdot x\text{H}_2\text{O}$ .

**Preparation of Organic Compounds:**

Preparation and techniques of purification of tribromophenol, nitrobenzene, aspirin, ethyl benzoate and benzoic acid from toluene, butyl chloride, acetanilide.

**Estimations (volumetric):**

- 1) Determination of molecular weight of a carboxylic acid.
- 2) Estimation of amide group and glucose.

**Recommended Books:**

1. Crocleford H.D. H.W. Biard F.W. Getzen & J.W. Nowell, "Laboratory Manual of Physical Chemistry" 2<sup>nd</sup> Edition John Wiley & Sons London.
2. Skoog, D.A., D.M. West and F.J. Holler, "Analytical Chemistry" 6<sup>th</sup> Ed. Saunders College Publications (1994).
3. Javed Iqbal Amin, "Theory and Practice of Chromatography", Higher Education Commission, Islamabad. (2002).
4. James A. Moore, "Experimental methods in Organic Chemistry", Holt-Saunders Int., 1983.
5. R.L. Shriner, R.C., Fuson D.V. Curtin and T.C. Morrill "The systematic identification of organic compounds, 6<sup>th</sup> Ed. John Willey & Sons 1979.



Code	Subject Title	Cr. Hrs	Semester
CHEM-301	Physical Chemistry	4	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Electrochemistry:**

Idea of conductance of electrolytes, Deby-Huckle equation and limiting law, ionic strength, weak electrolyte and Deby-Huckle theory, Activity and activity coefficients of electrolytic solution, determination of activities, concentration cells, Types of concentration cells, derivation of E.M.F of concentration cells with and without transference, Fuel cells and hydrocarbon cells.

**2. Quantum Chemistry:**

Postulates of quantum theory, Eigen functions, operators, Schrödinger's wave equation, particle in one dimensional box, Normalized wave function and orthogonality, Quantum mechanical tunneling, motion of particle in three dimensional box and idea of degeneracy, separation of variables and derivation of quantum numbers, Mathematical treatment of rigid rotator and calculation of bond length of simple molecules, harmonic oscillator and calculation of bond length of simple molecules, harmonic oscillator and calculation of vibrational frequencies, formation of covalent bond, Mathematical treatment of  $\text{He}_2^+$  and  $\text{H}_2$  molecules, discussion of overlapping integrals, molecular orbital theory and formation of  $\text{H}_2$  and  $\text{O}_2$  molecules.

**3. Chemical kinetics:**

Concept of rate law and order of reaction, Kinetics of 3rd order reaction with different concentrations and molecular identity, kinetics of opposing, parallel and consecutive reactions, basic experimental methods, Kinetics of thermally excited chain reactions like reaction of  $\text{H}_2$  and  $\text{Br}_2$ , kinetics of thermal decomposition of ozone,  $\text{N}_2\text{O}_5$  and  $\text{CH}_3\text{CHO}$ .

**RECOMMENDED BOOKS:**

1. Physical Chemistry, Samuel Glasstone, 1995. Macmillan and Co. Ltd. St. marlins Street, London.
  2. Principles of Physical chemistry, Maron and Prutton, 1965 the Macmillan Company, Collier Macmillan Ltd. London.
  3. Physical Chemistry, Barrow, 1973, McGraw Hill, Tokyo.
  4. Physical Chemistry, Moore, 1972, Rentice Hall, Englewood cliffs, Jersey.
  5. Physical Chemistry, Alberty and Daniels, 1962, McGraw Hill Book Company Ltd London.
  6. Physical chemistry, Alkins, 1989, Oxford University Press, Walton Street, Oxford.
- Physical Chemistry, Castallan, 1972, Addison Westey Publishing Company, Menla Park, California, London.





Code	Subject Title	Cr. Hrs	Semester
CHEM-302	Physical Chemistry Lab	2	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Basic Concepts:**

Preparation of standard molar and Normal solutions and percentage compositions of different compounds

**2. Chemical Kinetics:**

To investigate the kinetics of hydrolysis of ethyl in the presence of an acid.  
To determine the relative strength of acids (HCl and H<sub>2</sub>SO<sub>4</sub>) studying the hydrolysis of an ester.

**3. Electric conductance of electrolytes:**

To determine the cell constant of given cell.  
To determine the equivalence conductance of solution of weak electrolyte.  
At a no. of dilution at room temperature and from this result to verify Oswald's law.  
To determine the solubility of sparingly soluble salt.  
To determine the solubility of weak base of NH<sub>4</sub> OH by titrating it against Standard solution of HCl by using conductivity method.  
To determine the strength of given base by titrating it against standard Acetic acid solution and HCl solution using conductivity meter.  
To determine the strength of HCl and CH<sub>3</sub> COOH in the given mixture of both by titrating it against NaOH conductometrically.  
To determine the equivalent conductance of a weak electrolyte at infinite dilution using Kohlraush law.

**4. Phase Equilibria:**

To determine the partition coefficient of benzoic acid and iodine between CCl<sub>4</sub> and H<sub>2</sub>O.

**RECOMMENDED BOOKS:**

1. Advanced Experimental Physical Chemistry by Ayodhya Sing.
2. Experimental Physical Chemistry by Daniel
3. Experimental Physical Chemistry by G.Peter Matthews.
4. Experiments in Physical Chemistry by Shoemaker.



Code	Subject Title	Cr. Hrs	Semester
CHEM-303	Inorganic Chemistry	4	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Pi- acceptor Ligands:**

Transition metal carbonyls (Mononuclear, Binuclear, Polynuclear), synthesis, bonding situation based on spectroscopic evidences; Theoretical rationalization of molecular structures, (close, nido, erachno), Synthesis. Characteristics and reactivity of derivatives of metal carbonyls (carbonylate anions, carbonyl hydrides and carbonyl halides); Metal nitrosyls including halonitrosyl and their derivatives.

**2. Chemical bonding:**

Metallic bond on the basis of band model, X-ray spectra and  $N(E)$  curves,  $n(E)$  curves. Binding energy in metals, conductors, semi-conductors and insulators. Effect of temperature and impurities on conductivity.

**3. Coordination Compounds: (Structure & Bonding)**

Development of coordination compounds, Rules of inorganic nomenclature for acids, salts, radicals, ions, iso and heteropoly anions & compounds. Hybridization in coordination compounds with coordination number from 2 to 9. MO diagrams for metal complexes of common geometry. Important features of CFT, d-orbitals splitting for various common geometries, measurement of  $10 Dq$ ., factors effecting  $10 Dq$ . CFSE, factors influencing magnitude of variation in lattice and hydration energy for ions of first transition series.

**RECOMMENDED BOOKS:**

1. J H Huheey, Inorganic Chemistry - Principles, structure and reactivity, Harper and Row Publisher, Inc. New York (2008)
2. 3) J. D. Lee, Concise Inorganic Chemistry, Elbs with Chapman and Hall, London
3. Chemical Bonds, and introduction to atomic and molecular structure by H.B. Gray 1973, W.A. Benjamin, Inc., London
4. Advanced Inorganic Chemistry F.A. Cotton and G.Wilkinson 6th Ed. 2001, Interscience, Publishers, London.
5. Coordination Compounds by S.F.A. Kettle, 1999, Nelson , (Nairobi Kenya).



Code	Subject Title	Cr. Hrs	Semester
CHEM-304	Inorganic Chemistry Lab	2	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Chromatographic Techniques:**

- (a) Separation of metal ions by paper chromatography and their identification with the help of locating agents and comparison of  $R_f$  values.
- (b) Separation of anions by paper chromatography and their identification.

**2. Aqueous acid-base Titrations:**

- (a) Estimation of  $\text{SO}_2$  and  $\text{SO}_3$  in air and discharged from an industrial process.
- (b) Estimation of  $\text{CO}_2$
- (c) Estimation of oxalic acid and  $\text{H}_2\text{SO}_4$  in a mixture.
- (d) Estimation of  $\text{H}_3\text{BO}_3$  and  $\text{NaH}_2\text{BO}_3$  in a mixture.
- (e) Determination of %age composition of a mixture containing  $\text{H}_3\text{BO}_3$  and  $\text{CH}_3\text{COOH}$ .

**3. Precipitation Titrations:**

Estimation of following anions with the help of adsorption indicators:

- (i) Chloride (ii) Bromide (iii) Sulphate (iv) Chloride and Iodide in a mixture.

**RECOMMENDED BOOKS:**

1. J. Bassett, R. C. Denny, G. H. Jeffery and J. Mendham, Vogel's Text Book of qualitative Inorganic Analysis, the English Language Book Society and Longman, New York, (2008)
2. Quantitative Analysis Chemistry, James S. Pritz, George H. Sehenk, 2001 Alby and Becon Inc. London.
3. Theory and practice of chromatography by Prof. Dr. Javed Iqbal (2002).
4. Instrumental analysis by Gary D. Christian and James E.O., Reilly, 2007, Allyn and bacon Inc., London.
5. Hand Book of Organic reagents in Inorganic Analysis by ZAVIX Holzbecher and other 1976 Ellis Hurwod Limited, London.



Code	Subject Title	Cr. Hrs	Semester
CHEM-305	Organic Chemistry	4	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Acid-base strength:**

pKa and Ka values, electronic effects (Inductive and resonance effects), field effect, solvent effect, hyper-conjugation, hydrogen bonding, steric and stereo-chemical effects, and hybridization.

**2. Stereochemistry:****(a) Conformation Analysis**

The concept of conformational analysis in ethane, propane, n-butane, pentane, cyclopentane, cyclohexane, substituted alkanes, substituted cycloalkanes and decalins.

**(b) Optical isomerism:**

Configuration, Chirality and symmetry, optical isomerism upto three chiral carbon atoms, enantiomers and diastereomers, R and S nomenclature, Racemates, Racemization and Resolution of Racemates, epimerization. Walden inversion, Stereoisomerism in biphenyls, allenes and spiro-compounds

**(c) Geometrical isomerism**

Cis & Trans, and Z & E conventions, Determination of configuration, Geometrical isomerism in cyclic compounds.

**3. Active Methylene Compounds:**

Alkylation, arylation, and acylation of active methylene compounds, Acid and base catalysed aldol condensation. Conditions, mechanism and synthetic applications of the following reactions; Claisen, Claisen Schmidt, Knoevenagel, Perkin, Reformatsky, and Stobbes condensations, Darzen's glycosidic ester synthesis, Mannich and Wittig reactions.

**RECOMMENDED BOOKS:**

- Organic Chemistry, Volume I (6<sup>th</sup> ed.) & II (5<sup>th</sup> ed.) by I.L. Finar, Pearson Education (singapore) Pte Ltd, 2008.
- March's Advanced Organic Chemistry: Reactions, Mechanisms, and Structure, 6th ed. by Michael B. Smith, Jerry March, Wiley, 2007
- Organic Chemistry, 5th ed.; by S. H. Pine, McGraw Hill: New York, 1987
- Organic Chemistry 6th ed. by Francis A. Carey, McGraw Hill, 2005.
- Organic Chemistry 6<sup>th</sup> ed, by R. T. Morrison, R. N. Boyd, and R. K. Boyd, Benjamin Cummings, 1992,.
- Modern Synthetic Reactions 2nd ed. by H.O. House, W.A. Benjamin Inc., Menlo Park, CA
- Principles in Organic Synthesis by R.O.C Norman & J. M. Coxon, Chapman and Hall, 1993
- Organic Chemistry by Jonathan Clayden, Nick Geves, Stuart Warren, Oxford University Press 2000.



---

Code	Subject Title	Cr. Hrs	Semester
CHEM-306	Organic Chemistry Lab	2	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Organic Preparations:**

Benzyl alcohol; Ethyl benzene; benzoic acid, p-Nitrophenol, acetophenone oxime, acetophenone arylHydrazone.

**2. Quantitative and Qualitative Analysis of Organic compounds:**

Estimation of glucose, and Number of acetyl groups, two component mixture analysis.

**RECOMMENDED BOOKS:**

1. Practical Organic Chemistry by F. G. Mann and B. C. Saunders, Longman, UK, 1978.
  2. Vogel's Textbook of Practical Organic Chemistry (5th ed.) by A.I. Vogel, A.R. Tatchell, B.S. Furnis, A.J. Hannaford, P.W.G. Smith, 1989, Longman UK, 1989.
  3. The Systematic Identification of Organic Compounds, (8th ed.) by Ralph L. Shriner et al., Wiley, 2003.
  4. Advanced Practical Organic Chemistry, by J. Leonard, B. Lygo, G. Procter, CRC, 1994.
  5. Advanced Practical Organic Chemistry (2nd ed.) by N. K. Vishnoi, Vikas Publishing House Pvt Ltd, India, 1996.
-



Code	Subject Title	Cr. Hrs	Semester
CHEM-307	Analytical Chemistry	4	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Introduction / Assessment of Analytical Data**

Introduction and scope of Analytical Chemistry: Analytical problems and their solutions; The nature of analytical methods; trends in analytical methods; Different units of concentration and their conversion; Definition and basic concepts: nature and origin of errors, Classification of errors; Accuracy and Precision; Limits of detection, Confidence limits; Deviation, Standard deviation, Application of statistical tests; Rounding off analytical data; Quality control charts; Computation of analytical data. Significance of sampling, weighing and measuring in Analytical chemistry.

**2. Basic Chromatography Techniques**

Classifications of Chromatographic Techniques, Paper and Thin Layer Chromatographic Techniques; their instrumentation, applications and limitations, Column Adsorption Chromatography,

**3. Introduction to Spectroscopy / Spectrophotometry**

Introduction to Molecular spectroscopy, absorption in UV and Visible range; Basic principle of Spectrophotometry; Beer-Lambert's law; Deviations; Instrumentation and application.

**RECOMMENDED BOOKS:**

1. Analytical Chemistry by J.D. Dick, McGraw Hill, 1973, N.Y. also available in International students edition McGraw Hill, Mogakusha, 1973.
2. Instrumental Methods by W.Ewing, Mc Graw Hill Book Co. N.Y. (Third/Fourth Edition) also available in International students edition.
3. Chromatography by R.K Sharma , Gogel publishing home meerret
4. Introduction to chromatography by Nasir-ud-din, Published by author
5. Paper chromatography by Dr.Friedrich Cramer, London Macmillan and Co Ltd
6. Thin- layer chromatography by Marini, Elsevier publisher
7. Modern analytical chemistry by David Harvey, Roohani-art press, Islamabad
8. Principle and Practice of analytical chemistry by Fillfield, Blackwell Science Ltd
9. Spectroscopy by Browing, Mcgram Hill London
10. Fundamentals of Chromatography by H.G. Cassidy, Inter Science Publisher, London, N.Y.
11. Fundamentals of Analytical Chemistry by Doughlas Skoog and Donals M. W. West, Holt Reinchart and Inc, London.



Code	Subject Title	Cr. Hrs	Semester
CHEM-308	Analytical Chemistry Lab	2	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Calibration**

Calibration of glassware (pipette, burette and flask) used for volumetric analysis. Use of analytical balance and calculation of standard deviation. Calibration of pH meter and determination of pH of various acidic and basic solution.

Calibration of conductometer and determination of conductance of tap water, distilled water, conductivity water and canal water.

Calculation of dissociation constants of various acids.

Calculation of variance, mean, median, coefficient of variance of the data.

**2. Basic Chromatography**

Separation of ink components by paper chromatography.

Separation of amino acids by thin layer chromatography.

Separation of dyes by column chromatography.

Separation of mixtures by circular paper chromatography.

**RECOMMENDED BOOKS:**

1. Vogel's text book of quantitative inorganic analysis by J. Bassett. The English language book Society and Longman.
2. Introduction to chromatography by Nasir-ud-din, Published by author.
3. Paper chromatography by Dr. Friedrich Cramer, London Macmillan and Co Ltd.
4. Thin-layer chromatography by Marini, Elsevier publisher.



Code	Subject Title	Cr. Hrs	Semester
CHEM-309	Applied Chemistry	4	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Chemical Industrial Unit Operations:**

Brief introduction to Chemical Industry with reference to Pakistan, Elementary treatment of general unit operations commonly used in Industry such as heat transfer; Evaporation; size Reduction; Screening; Filtration and Distillation.

**2. Industrial Inorganic Synthesis:**

Raw materials; Chemical processes involved; flow sheet diagrams with all the important parameters concerned with the manufacturing of Sulphuric acid; Hydrochloric acid; caustic Soda; Washing soda; Applications of these chemicals in industry.

**3. Cement Industries:**

Cement raw materials used for cement manufacturing, dry process, wet process, semi wet process, special cement, chemistry involved in hydration of cement, setting of cement, setting time.

**4. Water Softening and Scale Removing:**

Water hardness; its measurement and removal; methods used for water softening including ion-exchange and reverse osmosis, distillation and precipitation. Types of boiler scales. Chemical and mechanical methods to eliminate the scaling.

**RECOMMENDED BOOKS:**

1. Industrial Organic Chemicals, by H.A.Witcoff and B.J.Reuben, John Wiley & Sons Inc. New York.
2. Water Supply and Sewerage, T.J.McGhee, McGraw Hill Book Co. New York.(1991)
3. Unit operations in Chemical Engineering, Chattopadhyay, Khanna Publishers, Delhi-6.(1993).
4. Chemical Process Design, Robin Smith, McGraw Hill Book Co. New York.(1995).
5. Hand Book of Industrial Chemicals, By SIRI Board of Consultants and Engineers,
6. Small Industries Research Institute, New Delhi (1995)
7. Small Medium and large Scale Industries, A.K. Sirivastawa, Small Industries Research Institute, New Delhi (1996).
8. The Chemistry of Cement, H.F.W. Taylor, Academic Press, London, 1964.
9. Shereve's Chemical Process Industries, 5th Ed.1975 by G.T.Austin McGraw Hill Book Co. New York.
10. Industrial chemistry, B. K. Sharma Krishna Prakashan Media (P) Ltd., Ed-15 (2006)





Code	Subject Title	Cr. Hrs	Semester
CHEM-310	Applied Chemistry Lab	2	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Preparations:**

Detergent and cosmetics (Cold cream, shampoo and vanishing cream)

**2. Titrimetry:**

Estimation of water hardness by complexometry

Estimation of acetic acid contents in the vinegar sample

Determine the acidity of the sulphuric acid and its normality.

Determination of acidity, alkalinity, Free CO<sub>2</sub> in water

Assay of bleaching powder by free chlorine method.

Determine the %age purity of the Commercial sample of sodium chloride.

**3. Flamephotometry:**

Estimation of Potassium in the tap water.

Estimation of Sodium in the Commercial Sodium Chloride.

Estimation of Calcium in milk.

**4. Spectrophotometry:**

Determination of the of KMnO<sub>4</sub>, K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> and CoCl<sub>4</sub>

**5. Chromatography:**

Separation of mixture of ink by circular paper chromatography.

Separation of mixture of metal ions by paper chromatography.

Coating of TLC plates and separation of mixture of dyes.

Separation of different pigments of plant extract by TLC chromatography.

**RECOMMENDED BOOKS:**

1. Perfumes Cosmetics and Soaps, W.A. Poucher, Chapman and Hall 7th Ed. (1974).
2. Applied Chemistry Theory and Practice O.P. Vermani & A.K. Narula, Wiley Eastern Limited (1989).
3. T. B. of Quantitative Inorganic Analysis, Vogel's Ed-4<sup>th</sup>, Longman Group Limited (1978).
4. Practical Statistics for the Analytical Scientist, A Bench Guide, RSC Publishing, LGC Ltd 2009



Code	Subject Title	Cr. Hrs	Semester
CHEM-311	Bio Chemistry	4	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Introductory Biochemistry:**

Scope of Biochemistry. The molecular logic of life. Structure and Functions of Cells. Cell wall Composition. A brief description on the isolation of cellular components.

**2. Water:**

Weak interactions in aqueous system. Ionization of water. Weak acids and weak bases. pH and buffer systems. Different buffering agents. Importance of buffers in biological systems.

**3. Carbohydrates:**

Nature, Structure and Classification of Carbohydrates. Aldoses and Ketoses Cyclic structure of monosaccharides Hawarth configurations, D and L configuration of monosaccharides, Optical isomerism and Mutarotation in glucose. Formation of Glycosidic bonds. Reducing and non reducing sugars. Important monosaccharide and their derivatives. Invert sugars. Biological significance of Glucose. Structures and functions of common Disaccharides and Polysaccharides: Succrose, Lactose, Maltose Amylose and Amylopectins. , Cellulose, Chitin, Glycogen, Starch and Dextran. Derived carbohydrates and hexose derivatives present in microorganisms. Sensory properties of monosaccharides. Proteoglycan and glycoproteins: their Structure and function. Hyluronic acid Chondroitin Sulphate and related compounds.

**4. Nucleic acids:**

Purines, Pyrimidines and nucleotides. Structure and functions of DNA, different type of RNA. Nucleic acid hydrolysis. Determination of Primary structure of Nucleic acids. Chemical synthesis of oligonucleotides.

**RECOMMENDED BOOKS:**

1. Principles of Biochemistry by Lehninger AL, Nelson DL and Cox MN, 2000 Pub: worth Publishers
2. Biochemistry by Lubert Stryer 2006 Pub: Freeman and Company
3. Biochemistry by Voet, and Pratt, 2004, John wiley and sons Inc.
4. Lippincott's Biochemistry by Champe.P C; Harvey. R. A and Ferrier. D. R. 3rd ed., 2004 Pub: J. b. Lippincott Company
5. Harpers Biochemistry, 27th ed. 2006 Pub: McGraw Hill Inc.



---

Code	Subject Title	Cr. Hrs	Semester
CHEM-312	Bio Chemistry Lab	2	V
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Carbohydrates:**

Qualitative tests for Carbohydrates. Distinction between pentoses and hexoses, aldoses and Ketoses, reducing and non reducing sugars mono and polysaccharides. Chromatography of sugars. Preparation of glycogen from liver. Acid and enzymic hydrolysis of glycogen.

**2. Nucleic Acids:**

Isolation of RNA from beef liver. Isolation of DNA from Calf Spleen. UV absorption of nucleic acids.

**RECOMMENDED BOOKS:**

1. Practical clinical Biochemistry by Varley. Pub: CBS publishers
  2. An Introduction to Practical Biochemistry By D. T. Plummer Pub: McGraw Hill
-



Code	Subject Title	Cr. Hrs	Semester
CHEM-313	Physical Chemistry	4	VI
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Kinetics of bimolecular reactions:**

Mathematical treatment of collision and transition state theory of bimolecular reactions, effect of temperature on reaction rates, the interpretation of bimolecular reactions in solution, ionic reaction in solution, unimolecular gas phase reactions, fast reactions and their methods of study.

**2. Classical Thermodynamics:**

Maxwell's relations and thermodynamics formula, second law of thermodynamics, Clausius inequality, the entropy of non ideality of a gas, Nerst heat theorem, its applications to solid and gases, Nerst approximation formula, third law of thermodynamics and determination of entropy by third law, Experimental verification of third law. Adiabatic demagnetization.

**3. Statistical Thermodynamics:**

Sterling's approximation, statistical treatment of entropy, partition function and its physical significance, absolute entropy and partition functions, interpretation of thermodynamic functions in terms of translational, rotational vibrational and electronic partition functions, Free energy and equilibrium constant from partition function.

**4. Kinetic theory of Gases:**

Introduction, Maxwell distribution of molecular velocities and energies, Derivation of average velocity and most probable velocity, Barometric formula, effect of altitude, molar mass and temperature on vertical distribution of particles.

**Note Suggested out lines for Mathematics Course:****Basic Mathematics for Chemistry (1 Credit Course):**

Review of basic algebra and trigonometry, concept of function, Differentiation, concept of maxima and minima, integration, Differential Equations, equations of straight lines, partial fraction, Applications of Calculus in Chemistry and data handling.

**RECOMMENDED BOOKS:**

1. Physical Chemistry by Kundu, N and Jain, S.K., S. Chand and Company Ltd. 1984.
2. Fundamentals of Chemical kinetics by Logan, S.R., Longman Group Ltd. 1996.
3. Elementary reaction kinetics by Latham. J.L. and Burgess, A.E., 3rd Ed., Butterworths, London, 1997.
4. Physical Chemistry by Atkins, P.W., 5th Ed., W.H. Freeman and Company, New York, 1994.
5. Physical Chemistry by Alberty, R.A. and Silbey, R.J., John Wiley, New York, 1995.
6. Physical Chemistry by Engel, T. and Ried, P., 1st Ed., Pearson education, Inc. 2006.



7. Electrochemical Methods and applications by bard, A. and Faulkner, L.R., John Wiley, New York, 1980.
  8. Principles of Physical Chemistry by Maron and Prutton, Macmillan and Co. Ltd. 1965.
  9. Physical Chemistry by Glasstone, S. Macmillan and Co. Ltd., London, 195.
  10. Elements of classical and statistical thermodynamics by Nash, L.K. Addison Wesley Co. Ltd., 1979.
  11. Mathematics for Chemists.
-



Code	Subject Title	Cr. Hrs	Semester
CHEM-314	Physical Chemistry Lab	2	VI
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Refractrometry:**

To determine the unknown concentration of sucrose solution and ethanol solution.

**2. Molar mass determination (Colligative properties):**

To determine the molecular weight of a substance by cryoscopic method and Ebullioscopic method.

**3. Spectrophotometry:**

To determine the wavelength of maximum absorption of compounds using spectrophotometer.

To determine the unknown concentration of a compound using spectrophotometer.

**4. Phase Equilibrium:**

To determine the phase diagram of Naphthalene and diphenyl system.

To determine the phase diagram of urea and phenol.

To determine the phase diagram of Benzoic acid and Naphthalene.

**5. Optical activity measurement:**

To determine the unknown percentage composition of the following by using polarimeter (Sucrose, glucose).

To determine the specific and molar rotation of optically active compound (sucrose, glucose).

**RECOMMENDED BOOKS:**

1. Advanced Experimental Physical Chemistry by Ayodhya Sing.
2. Experimental Physical Chemistry by Daniel
3. Experimental Physical Chemistry by G.Peter Matthews.
4. Experiments in Physical Chemistry by Shoemaker.



Code	Subject Title	Cr. Hrs	Semester
CHEM-315	Inorganic Chemistry	4	VI
Year	Discipline		
3	Chemistry-I, II		

### SYLLABUS OUTLINE:

#### 1. The Covalent Bond (Structure & Reactivity):

- VSEPR model followed by VB theory (Hybridization, Resonance etc.) explanation of the structure of  $AB_2$ ,  $AB_3$ ,  $AB_2E$ ,  $AB_4$ ,  $AB_3E$ ,  $AB_2E_2$ ,  $AB_5$ ,  $AB_3E_3$ ,  $AB_6$ ,  $AB_5E$ ,  $AB_4E_2$ ,  $AB_7$ ,  $AB_6E$ ,  $AB_8$  and  $AB_9$  type molecules.
- Discussion of molecular orbitals and molecular structures of homonuclear molecules and ions, heteronuclear diatomic and polyatomic molecules and ions.
- Bent bond, bridge bond, four electrons-three centre bond.
- Shielding effect and effective nuclear charge, Factors affecting the magnitude of  $\sigma$  and  $Z_{\text{eff}}$  and their variation in the period table, Applications of Slater's rules, Polarization of ions, Fajan's rules and its applications.

#### 2. Co-ordination compounds: (synthesis and properties)

Preparative methods. Techniques of studying complexes, stability constants. The spectrochemical series and colour of metal complexes. Diamagnetism and Paramagnetism, stereochemistry, John-Teller Theorem, Isomerism. Role of metal complexes in analytical chemistry, industry and nature.

#### 3. Chemistry of the Lanthanides and Actinides

Nomenclature, Position in periodic table, occurrence, Separation, and electronic configuration, oxidation States, Complex Formation, shapes of 'f'-orbitals, applications.

### RECOMMENDED BOOKS:

- J H Huheey, Inorganic Chemistry - Principles, structure and reactivity, Harper and Row Publisher, Inc. New York (2008)
- J. D. Lee, Concise Inorganic Chemistry, Elbs with Chapman and Hall, London (2007).
- Advanced Inorganic Chemistry F.A. Cotton and G.Wilkinson 6th Ed. 2001, Interscience, Publishers, London.
- Coordination Compounds by S.F.A. Kettle, 1999, Nelson , (Nairobi Kenya).
- Coordination Chemistry by B.A. Basallo and R. Johnson 1972 W.A. Benjamin, London.



Code	Subject Title	Cr. Hrs	Semester
CHEM-316	Inorganic Chemistry Lab	2	VI
Year	Discipline		
3	Chemistry-I, II		

### SYLLABUS OUTLINE:

#### 1. Complexometric Titrations:

- Estimation of  $Mg^{+2}$   $Zn^{+2}$  with EDTA (Direct titration).
- Estimation of  $Ni^{+2}$  with EDTA (Back titration).
- Determination of  $Ca^{+2}$  and  $Zn^{+2}$  in mixture (Masking)
- Determination of  $Cd^{2+}$  and  $Zn^{+2}$  in a mixture (Demasking).
- Determination of  $SO_4^{2-}$  and  $PO_4^{-3}$  with EDTA (Indirect titration).

#### 2. Redox Titrations:

- Use of Ceric sulphate solution for the estimation of the following:-
  - Determination of iron in an iron ore.
  - Determination of nitrites.
- Use of potassium iodate for the determination of the followings:
  - Copper
  - $H_2O_2$
  - Commercial Hypochlorite

#### 3. Preparations:

- Tris (ethylenediamine ) nickle (II) Chloride 2-hydrate.
- Sodium Cobaltinitrite.
- Pot. Trioxalato Aluminate.
- Ammonium sulphate Nickel (II) Sulphate.
- Hexa aqua Chromium (III) Chloride.
- Ammonium Sulphate Copper (II) Sulphate Pentahydrate.

### RECOMMENDED BOOKS:

- J. Bassett, R. C. Denny, G. H. Jeffery and J. Mendham, Vogel's Text Book of qualitative Inorganic Analysis, the English Language Book Society and Longman, New York, (2008)
- Quantitative Analysis Chemistry, James S. Pritz, George H. Sehenk, 2001 Alby and Becon Inc. London.
- Theory and practice of chromatography by Prof. Dr. Javed Iqbal (2002).
- Instrumental analysis by Gary D. Christian and James E.O., Reilly, 2007, Allyn and bacon Inc., London.
- Hand Book of Organic reagents in Inorganic Analysis by ZAVIX Holzbecher and other 1976 Ellis Hurwod Limited, London.
- Experimental Inorganic Chemistry - W. G. Palmer, 2005.
- The analysis of minerals and ores of the rarer elements – W. R. Schoeller, and A. R. Powell, Charles, Griffin and Company Limited, 2004.





Code	Subject Title	Cr. Hrs	Semester
CHEM-317	Organic Chemistry	4	VI
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Free radical Reactions:**

Introduction, generation methods, relative stability, structure, free radical reactions and industrial applications.

**2. Oxidation and Reduction reactions:****(a) Oxidation Reactions:**

Introduction, Oxidation of saturated hydrocarbons, olefinic double bonds, aromatic rings, systems containing oxygen such as alcohols, aldehydes, ketones, oxidative decarboxylation, of acids, oxidation of systems containing nitrogen such as amines, hydrazines etc..

**(b) Reduction Reactions:**

Introduction, Reduction of alkenes, alkynes, and aromatic rings, hydrogenolysis, reduction of benzylic and allylic systems, aldehydes and ketones, alcohols, pinacols, epoxides, acids and their derivatives, Reduction of system containing nitrogen such as imines, oximes and nitro compounds

**3. Spectroscopy:****(a) IR Spectroscopy:**

Electromagnetic radiations: IR; modes of vibration, sampling techniques, factors influencing the vibration frequencies and industrial applications

**(b) UV Spectroscopy:**

Ultraviolet (UV) or electronic spectroscopy: electronic transition; factors influencing the  $\lambda_{max}$ . value.

**RECOMMENDED BOOKS:**

- Organic Chemistry, Volume I (6th ed.) & II (5th ed.) by I.L. Finar, Pearson Education (Singapore) Pte Ltd, 2008.
- March's Advanced Organic Chemistry: Reactions, Mechanisms, and Structure, 6th ed. by Michael B. Smith, Jerry March, Wiley, 2007.
- Organic Chemistry, 6th ed.; by S. H. Pine, McGraw Hill: New York, 1987.
- Organic Chemistry 6th ed. by Francis A. Carey, McGraw Hill, 2005.
- Organic Chemistry 6th ed. by R. T. Morrison, R. N. Boyd, and R. K. Boyd, Benjamin Cummings, 1992.
- Modern Synthetic Reactions 2nd ed. by H.O. House, W.A. Benjamin Inc., Menlo Park, CA
- Principles in Organic Synthesis by R.O.C Norman & J. M. Coxon, 1993, Chapman and Hall, 1993.
- Organic Chemistry by Jonathan Clayden, Nick Greeves, Stuart Warren, Oxford University Press 2000.



9. Spectroscopic Methods in Organic Chemistry 6th ed. by D. Williams and I. Fleming. Wiley-VCH, 1991.
  10. Spectrometric identification of Organic Compounds 6th ed. by R. M. Silverstein and F. X. Webster, Wiley, 2007.
  11. Organic Spectroscopy and Chromatography by M Younas, ILMI, Pakistan, 2007.
-



---

Code	Subject Title	Cr. Hrs	Semester
CHEM-318	Organic Chemistry Lab	2	VI
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Organic Preparations:**

Synthesis of compounds containing nitro, halogeno, amino, carboxylic and carbonyl functionalities (depends upon the availability of chemicals).

**2. Quantitative and Qualitative Analysis of Organic compounds:**

Physical/ Chemical separation of mixture containing two compounds, identification and derivitization.

**RECOMMENDED BOOKS:**

1. Practical Organic Chemistry by F. G. Mann and B. C. Saunders, Longman, UK., 1978.
  2. Vogel's Textbook of Practical Organic Chemistry (5th ed.) by A.I. Vogel, A.R. Tatchell, B.S. Furnis, A.J. Hannaford, P.W.G. Smith, Longman UK, 1989.
  3. The Systematic Identification of Organic Compounds, (8th ed.) by Ralph L. Shriner et al., Wiley, 2003.
  4. Advanced Practical Organic Chemistry, by J. Leonard, B. Lygo, G. Procter, CRC., 1994.
  5. Advanced Practical Organic Chemistry (2nd ed.) by N. K. Vishnoi, Vikas Publishing House Pvt Ltd , India., 1996.
-



---

Code	Subject Title	Cr. Hrs	Semester
CHEM-319	Analytical Chemistry	4	VI
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****Ion Exchange Chromatography:**

Cation Exchange resin, Anion Exchange resin, Cross-linkage, Effect of pH-separation of Amino Acids, Separation of metal ions on Anions Exchange Columns, Applications of ion Exchange Chromatography.

**Solvent Extraction:**

Basic principle of solvent extraction, The Distribution Coefficient, The Distribution Ratio, The Percent Extracted Solvent Extraction of Metals, Analytical Separations, Multiple Batch Extractions, Countercurrent Distribution, Solid-Phase Extraction, Solvent Extraction by Flow Injection Analysis.

**Electrophoresis:**

Capillary Zone Electrophoresis, Application of traditional Electrophoresis Gel Chromatography.

**Flame Emission:**

Basic principle of atomic spectroscopy; Use of atomic spectra for detection and determination of elements; flame as a source of atomization and excitation; Instrumentation involved in FES; applications and limitations.

**Atomic Absorption Spectroscopy:**

Basic Principle of AAS; Flameless AA spectroscopy including graphite furnace and hydride generation.

**RECOMMENDED BOOKS:**

1. Vogels, text book of Quantitative chemical analysis by J. Mendham, R. C. Denney, J. D. Barnes, M. J. Thomas, Pearson Education Ltd.
  2. Advances in electrophoresis by Andrea Chrmambach, Wiley-VCH.
  3. Ion-Exchange Chromatography by Helfferich, McGraw Hill Book Co., Inc. N.Y. London.
  4. Solvent Extraction by Gorge H. & Morrison Hener, John Wiley and sons, London, N.Y.
  5. Chromatographic Methods of Analysis by Stock & Rice, Elsevier Co. Amsterdam.
  6. Flow injection analysis by Ruzicke Hassen, Wiley Interscience.
-



Code	Subject Title	Cr. Hrs	Semester
CHEM-320	Analytical Chemistry Lab	2	VI
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****Flame Emission / Spectrophotometry:**

Determination of Sodium in tap water by using Flame Photometer.  
 Determination of Potassium in tap water by using Flame Photometer.  
 Find out the calcium in chalk sample by flame photometry.  
 Determination of Ba by flame photometry.  
 Estimation of purity of various compounds on the base of flame emission Spectrophotometry.  
 Indirect determination of various compounds by flame photometric techniques.  
 Determination of  $\lambda_{max}$  of  $KMnO_4$  and  $K_2Cr_2O_7$  by using spectrophotometer.  
 Verification of Beer and Lambert Law.  
 Ultraviolet spectrophotometric determination of Aspirin, Phenacetin and Caffeine in APC tablet using Solvent Extraction.

**RECOMMENDED BOOKS:**

1. Vogel's text book of quantitative inorganic analysis by J. Bassett. The English language book Society and Longman
2. Flame spectrometry in environmental chemical analysis. A practical guide (RSc) analytical spectroscopy series by Malcolms Cresser alfreds sanz medel ,[http// www.Amazon.co.uk](http://www.Amazon.co.uk)



Code	Subject Title	Cr. Hrs	Semester
CHEM-321	Applied Chemistry	4	VI
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Chemical unit processes:**

Elementary treatment of general unit Chemical unit processes, Nitration; Halogenation; Sulphonation; Hydration and Hydrogenation and Oxidation with real examples from Industry.

**2. Industrial Organic Synthesis:**

Raw materials; Chemical processes involved; flow sheet diagrams with all the important parameters concerned with the manufacturing of Oxalic acid; Phthalic anhydride; Acetic acid; Phenol; Ethyl Acetate; Butadiene and Styrene

**3. Glass Industries:**

History of glass, raw materials used for glass, methods of manufacturing, various types of furnaces and crucibles used for the manufacture of glass, special types of glass, their manufacture and properties.

**4. Soap and Detergent Industries:**

Processes involved in soap manufacturing, methods used for manufacture of laundry soap, typical soaps. Recovery of glycerine. Detergents or surface active agents, cationic, anionic and non-ionic agents.

**RECOMMENDED BOOKS:**

1. Shereve's Chemical Process Industries, 5th Ed.1975 by G.T.Austin, McGraw Hill Book Co. New York.
2. Industrial Organic Chemicals, by H.A.Witcoff and B.J.Reuben, John Wiley & Sons Inc. New York.
3. Riegel's handbook of Industrial Chemistry, Ed. J.A.Kent, CBS Publishers and Distributors, New Delhi.(1997).
4. Chemical Process Design, Robin Smith, McGraw Hill Book Co. New York.(1995).
5. Hand Book of Industrial Chemicals, by SIRI Board of Consultants and Engineers, Small Industries Research Institute, New Delhi (1995).
6. Chemistry of glass manufacturing, F.W.Hunter, Dower Publications, New York, 1950.
7. Industrial chemistry, B. K. Sharma, Krishna Prakashan Media (P) Ltd., Ed-15 (2006).



Code	Subject Title	Cr. Hrs	Semester
CHEM-322	Applied Chemistry Lab	2	VI
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. Preparations:**

Dentrifrice, Thermosetting and thermoplastic resins (alkyd and urea formaldehyde)

**2. Chromatography:**

Coating of TLC plates and separation of mixture of dyes.

Separation of different pigments of plant extract by TLC chromatography Packing of chromatographic column and separation of a mixture of dyes. Separation of  $\beta$ -carotene from the pulp of tomato and carrot by column chromatography.

**3. Titrimetry:**

Analysis of soda ash and caustic soda in a mixture. Determine the active material in the commercial sample of detergent by cationic titration. Determining the %age purity of sodium bicarbonate and sodium carbonate. Determination of Residual Chlorine in water. Sage of reducing sugars. Soap analysis for free and combined alkali.

**4. Spectrophotometry:**

Estimation of nickel in vanaspati ghee.

Estimation of chloride in the tannery effluent.

Estimation of Iron in Pharmaceutical Products.

Estimation of Phosphates in fertilizers.

**RECOMMENDED BOOKS:**

1. Perfumes Cosmetics and Soaps, W.A. Poucher, Chapman and Hall 7th Ed. (1974).
2. Applied Chemistry Theory and Practice, O.P. Vermani & A.K. Narula, Wiley Eastern Limited (1989).
3. T. B. of Quantitative Inorganic Analysis, Vogel's Ed-4<sup>th</sup>, Longman Group Limited (1978).
4. Practical Statistics for the Analytical Scientist, A Bench Guide, RSC Publishing, LGC Ltd 2009.



Code	Subject Title	Cr. Hrs	Semester
CHEM-323	Bio Chemistry	4	VI
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:****1. LIPIDS:**

Lipid Classification, Structures and functions. Chemical Properties of triglycerides. Phospholipids. Sterol/steroids. Lipid with specific biological activities. Prostaglandins: Structure and function. Properties of lipid aggregates: Micelles and Bilayers. Biological membranes. Membrane proteins, Membrane structure and Assembly. Fluid Mosaic model. The erythrocyte membrane.

**2. PROTEINS:**

Amino acids: their Structure, Chiral Center, and stereoisomerism. Classification of amino acids. Acid base properties, their titration curve and its importance. . Amino acid sequence. Peptides and their biological importance. Proteins: classification, Covalent structure and biological significance including Primary. Secondary, Tertiary and Quaternary structure of proteins, as Keratins, Collagens and elastin. Conformation and function of globular proteins with special reference to structure and function of Hemoglobin and Myoglobin. Biological significance of Proteins.

**3. ENZYMES:**

Chemical nature, nomenclature and classification of enzymes. Cofactors and Coenzymes. Concepts of Active site. Substrate specificity. Affect of different factors on enzyme activity. Kinetics of single substrate reactions. Quantitative assay of enzymatic activity. Enzyme inhibition: Competitive, non-competitive and irreversible inhibition. Regulatory enzymes, allosteric enzymes, Multienzyme system, Zymogens, isoenzyme. Immobilized enzymes.

**4. NUTRITION:**

Introduction to the science of nutrition: Nutrients and their functions Biological evaluation of proteins, carbohydrates and lipids. Sources and forms of Energy. Energy value of foods. Energy requirements under different living and physiological conditions. Direct and indirect Calorimetry. Basal metabolic Rate, Respiratory quotient and their measurements. Assessment of nutritional status in Pakistan. Thermogenic effects of food.

**RECOMMENDED BOOKS:**

1. Principles of Biochemistry by Lehninger AL, Nelson DL and Cox MN, 2000 Pub: worth Publishers
2. Biochemistry by Lubert Stryer(2006) Pub: Freeman and Company
3. Lippincott's Biochemistry by Champe.P C; Harvey. R. A and Ferrier. D. R. 3rd ed., 2004 Pub: J. b. Lippincott Company
4. Nutritional biochemistry by T. Brody 1994, Pub: Acadmic Press
5. Harpers Biochemistry, 27th ed. 2006 Pub: McGraw Hill Inc.





Code	Subject Title	Cr. Hrs	Semester
CHEM-324	Bio Chemistry Lab	2	VI
Year	Discipline		
3	Chemistry-I, II		

**SYLLABUS OUTLINE:**

**1. Lipids:**

Qualitative tests for lipids including fatty acids, sterols and phospholipids. Lipids separation from Calf brain tissue. Acid value, Saponification Value and Iodine Value of fats. Extraction and TLC of Wheat Lipids.

**2. Amino Acids and Proteins:**

Qualitative tests of amino acids, determination of isoelectric Point. Isolation and solubilization of proteins from plant and animal origin. Hydrolysis of proteins. Estimation of proteins by Kjeldahl method. Isolation of enzyme amylase, a study on its properties and catalytic activity.

**RECOMMENDED BOOKS:**

1. Practical clinical Biochemistry by Varley. Pub: CBS publisher
2. An Introduction to Practical Biochemistry By D. T. Plummer 3rd ed. (1987) Pub: McGraw Hill

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-401	Environmental Chemistry	2	VII
Year	Discipline		
4	Chemistry		

### **SYLLABUS OUTLINE:**

#### **1. Introduction:**

History and significance of environmental degradation, impact of the modern life-style on environmental quality, resource dep} at environmental conservation and sustainability poverty and environmental degradation, environmental education , institutions for the protection of environment, inter-disc nature of environmental studies, environmental segments.

#### **2. Atmospheric pollution:**

Importance of air, nature and composition of atmosphere, temperature and pressure profi of different layers of the atmosphere, common air pollutants and their sources, oxides of C, N, and S hydrologic cycle, green house effect and global warming, stratospheric ozone depletion, vehicular emissions, particulate matter and aerosols, airborne lead, acid rain and its impats, photochemical smog, photochemistry of the atmosphere, role of hydroradicals, indoor air quality.

#### **3. Water pollution and water treatment:**

Importance of water, physical and chem. Properties of water, criteria for water quality, BOD and COD, sources of water pollution (industrial, agricultural, municipal and natural), fertilizers, pesticides, detergents, heav metals, bio-accumulation and bio-impliflcat primary, secondary and advanced treatment of water.

### **RECOMMENDED BOOKS:**

1. Kumar. Environmental Chemistry, Wiley Eastern, New Delhi.
2. J.W. Moore & EM. Moore, Environmental Chemistry, Academic Press, New York.
3. S. K. Banerji, Environmental Chemistry, Prentice Hafl, Delhi.
4. K. Banerji, Environmental Chemistry, Tata Publisher, Delhi.
5. Staneley E. Manahan, Environmental Chemistry, Brooks, California.
6. Neil, P.O. Environmental Chemistry, Chapmann, London.
7. Baird, C. Environmental Chemistry, Freeman, New York.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-402</b>	<b>Environmental Chemistry</b>	<b>2</b>	<b>VIII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### **SYLLABUS OUTLINE:**

**1. Land pollution:**

Importance of soil, nature and composition of soil, macro and micro-nutrients in soil, soil erosion, pH of soil and nutrients availability, on-exchange in soil, sources of soil pollution (industry, agrochemicals, mining, municipal waste, littering), reclamation of soil.

**2. Chemicals in the Environment and their impact:**

Chemical speciation, heavy metals, persistent organic pollutants, aflatoxins, PCB's, pesticides and detergents, house hold chemicals, solvents. Impact of chemicals on human health, crops and vegetation, buildings and monuments, aquatic life, biodiversity, visibility, concept of green chemistry.

**3. Monitoring of Environmental Pollution and Legislation:**

Principle, applications of analytical techniques for monitoring of pollution with special reference to OC, HPLC. – UV and IR spectrometry, Atomic absorption spectroscopy. Legislation aspects of environmental pollution, international standards regarding environmental pollution.

### **RECOMMENDED BOOKS:**

1. Kumar. Environmental Chemistry, Wiley Eastern, New Delhi.
2. J.W. Moore & EM. Moore, Environmental Chemistry, Academic Press, New York.
3. S. K. Banerji, Environmental Chemistry, Prentice Hall, Delhi.
4. K. Banerji, Environmental Chemistry, Tata Publisher, Delhi.
5. Staneley E. Manahan, Environmental Chemistry, Brooks, California.
6. Neil, P.O. Environmental Chemistry, Chapman, London.
7. Baird, C. Environmental Chemistry, Freeman, New York.



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-403</b>	<b>Physical Chemistry (Sp. Theory-I)</b>	<b>4</b>	<b>VII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

**SYLLABUS OUTLINE:****1. Colloids and Surfactants:**

Colloids, Colloidal dispersions, sols and their preparation, properties of suspensions, Optional properties of sols, determination of particle size, kinetic properties of sols, sedimentation of suspensions, electrical properties of sols, electrophoresis and electro osmosis, stability of suspensions, precipitation of sols, associated colloids, macromolecular properties in solutions and molecular weight determinations.

**2. Advance approach to homogenous and heterogeneous kinetics:**

Adsorption isotherms, single system, double system, study of gas reactions on solid surfaces, retardation, the Eley-Rideal mechanism and the Langmuir-Hinshelwood mechanism to study some organic and inorganic reactions, Catalysis, Autocatalysis, enzyme catalysis and enzyme inhibition.

**RECOMMENDED BOOKS:**

1. Physical Chemistry by Kundu, N and Jain, S.K.S. Chand and Company Ltd. 1984.
2. Fundamentals of chemical kinetics by Logan, S.R, Longman Group Ltd. 1996.
3. Elementary reaction kinetics by Latham.J.L. And Burgess, A.E.3rd Ed., Butterworths, London, 1977.
4. Physical chemistry by Atkins, P.W. 5th Ed., W.H.Freeman and Company, New York, 1994.
5. Physical Chemistry by Alberty, R.A. and Silbey. R.J., John Wiley, New York, 1995.
6. Physical chemistry by Engel, T. and Ried, P., 1st Ed., Pearson Education, Inc. 2006.
7. Hand book of surface and Colloid Chemistry by Birdi, K.S., CRC Press, 1997.
8. Heterogeneous Catalysis: Principles and applications by Bond, G.C., 2nd Ed., Oxford, Clarendon press, 1987.
9. Surfactants and interfacial Phenomena by Rosen, Milton J., John Wiley, New York, 1978.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM- 404</b>	<b>Physical Chemistry (Sp. Theory-II)</b>	<b>4</b>	<b>VII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### **SYLLABUS OUTLINE:**

#### **1. Rotational and Vibrational Spectroscopy**

Special regions and classification of spectroscopy; Rotational energies of diatomic molecules, population of Rotational energy level. Rotational spectra of rigid linear molecules and determination of bond lengths. The Stark effect.

Vibrational spectroscopy: energy of an atomic molecule, harmonic and harmonic oscillator molecules, relative population of energy levels and intensities of transition, types of vibrational modes.

Vibrational of polyatomic molecules, interpretation of IR spectra of simple molecules, Fermi resonance, applications and sampling techniques.

#### **2. Photo Chemistry:**

Laws of photochemistry, quantum efficiency and its determination Photochemical reactions, excited state symbols; photosensitized reactions, phosphorescence, fluorescence, chemiluminescence, Lasers.

#### **3. Advanced Treatment of Solutions:**

The thermodynamic properties of solution. The solution process. Conditions of equilibrium between phases. Theoretical basis of Raoult's equation. Deviation from ideal behavior. Compound formation and association. Separation of solid solutions.

### **RECOMMENDED BOOKS:**

1. Physical Chemistry by Kundu, N and Jain, S.K.S. Chand and Company Ltd. 1984.
2. Fundamentals of chemical kinetics by Logan, S.R, Longman Group Ltd. 1996.
3. Elementary reaction kinetics by Latham.J.L. And Burgess, A.E.3rd Ed., Butterworths, London, 1977.
4. Physical chemistry by Atkins, P.W. 5th Ed., W.H.Freeman and Company, New York, 1994.
5. Physical Chemistry by Alberty, R.A. and Silbey. R.J., John Wiley, New York, 1995.
6. Physical chemistry by Engel, T. and Ried, P., 1st Ed., Pearson Education, Inc. 2006.
7. Hand book of surface and Colloid Chemistry by Birdi, K.S., CRC Press, 1997.
8. Heterogeneous Catalysis: Principles and applications by Bond, G.C., 2nd Ed., Oxford, Clarendon press, 1987.
9. Surfactants and interfacial Phenomena by Rosen, Milton J., John Wiley, New York, 1978.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-405</b>	<b>Physical Chemistry (Sp. Practical)</b>	<b>2</b>	<b>VII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### **SYLLABUS OUTLINE:**

- 1. Molecular Mass Determination of different Polymers by Viscosity measurement.**
- 2. Thermo-chemical measurements**
  - Heat of solution of a salt by solubility methods.
  - Heat of precipitation.
- 3. Colloidal State:**

Preparation of As<sub>2</sub>S<sub>3</sub> sol, and comparison of precipitating powers of different cations.
- 4. Spectrophotometry:**
  - Determination of percentage composition of two coloured components in solution.
  - Study of kinetics of iodination of acetone and decomposition of benzene diazonium chloride.
  - Predicting normal modes of vibration for simple molecules and interpretation of their IR Spectra.

### **RECOMMENDED BOOKS:**

- Advanced Experimental Physical Chemistry by Ayodhya Sing.
- Experimental Physical Chemistry by Daniel
- Experimental Physical Chemistry by G.Peter Matthews.
- Experiments in Physical Chemistry by Shoemaker

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-422	Physical Chemistry (Sp. Theory-I)	4	VIII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### 1. Nuclear Chemistry:

Composition of the nucleus, natural and artificial radioactivity, rate of radioactive disintegration, radioactive equilibrium, transformation of elements cyclotron and linear accelerators; nuclear processes; nuclear fission, atomic bomb, nuclear reactor, nuclear fusion, hydrogen bomb, stellar energy, radiation hazards, use of tracers in chemistry.

#### 2. Advance Approach to Osmosis and Osmotic Pressure:

Semi Permeable membranes. The cause of semi-permeability. Mechanism of osmotic pressure. Dilute solutions and the Gas Laws. The Bombardment theory. Objections to the Bombardment theory. Review of the theories. Determination of the molecular weight by Osmometry.

#### 3. Gels and Emulsions:

Introductions, Methods of Preparation of Emulsions. Emulsifiers, Breaking of emulsions. Orientation Theory. Emulsification and wetting, Significance.

### **RECOMMENDED BOOKS:**

1. Physical Chemistry by Kundu, N and Jain, S.K.S. Chand and Company Ltd. 1984.
2. Fundamentals of chemical kinetics by Logan, S.R, Longman Group Ltd. 1996.
3. Elementary reaction kinetics by Latham.J.L. And Burgess, A.E.3rd Ed., Butterworths, London, 1977.
4. Physical chemistry by Atkins, P.W. 5th Ed., W.H.Freeman and Company, New York, 1994.
5. Physical Chemistry by Alberty, R.A. and Silbey. R.J., John Wiley, New York, 1995.
6. Physical chemistry by Engel, T. and Ried, P., 1st Ed., Pearson Education, Inc. 2006.
7. Hand book of surface and Colloid Chemistry by Birdi, K.S., CRC Press, 1997.
8. Heterogeneous Catalysis: Principles and applications by Bond, G.C., 2nd Ed., Oxford, Clarendon press, 1987.
9. Surfactants and interfacial Phenomena by Rosen, Milton J., John Wiley, New York, 1978.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-423	Physical Chemistry (Sp. Theory-II)	4	VIII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### 1. **Polymers:**

Classification of polymers; kinetics of condensation, addition and co-polymerisation reactions; Molecular mass distribution, determination of molecular masses by different methods. Analysis techniques.

#### 2. **Electronic and Raman Spectroscopy:**

Principles of electronic transition. Types of electronic transition. Energies of atomic orbital-with reference of H-atom spectrum electronic angular momentum fine structure of H-atom spectrum.

Raman Spectra-idea of Raman scattering, Rayleigh scattering Molecular polarizability. Rotational Raman Spectra of linear Molecules. Symmetric top molecules and spherical top molecules Vibrational Raman spectra.

### **RECOMMENDED BOOKS:**

1. Physical Chemistry by Kundu, N and Jain, S.K.S. Chand and Company Ltd. 1984.
2. Fundamentals of chemical kinetics by Logan, S.R, Longman Group Ltd. 1996.
3. Elementary reaction kinetics by Latham.J.L. And Burgess, A.E.3rd Ed., Butterworths, London, 1977.
4. Physical chemistry by Atkins, P.W. 5th Ed., W.H.Freeman and Company, New York, 1994.
5. Physical Chemistry by Alberty, R.A. and Silbey. R.J., John Wiley, New York, 1995.
6. Physical chemistry by Engel, T. and Ried, P., 1st Ed., Pearson Education, Inc. 2006.
7. Hand book of surface and Colloid Chemistry by Birdi, K.S., CRC Press, 1997.
8. Heterogeneous Catalysis: Principles and applications by Bond, G.C., 2nd Ed., Oxford, Clarendon press, 1987.
9. Surfactants and interfacial Phenomena by Rosen, Milton J., John Wiley, New York, 1978.



## BS (4 Years) for Affiliated Colleges



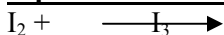
Code	Subject Title	Cr. Hrs	Semester
CHEM-424	Physical Chemistry (Sp. Practical)	2	VIII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

1. **Polarimetric Study of Reaction Kinetics:**

Inversion of cane sugar; salt effects on inversion of cane sugar.

2. **Equilibrium constant for the reaction:**



Calculation of  $\Delta G$  for the formation of  $I_3^-$  reaction.

3. **Adsorption measurements:**

Verification of Freundlich and Langmuir adsorption isotherms for Adsorption of acetic acid on active charcoal

4. **Potentiometric measurements:**

A. Potentiometric Titrations

(i) Acid base (ii) Oxidation-reduction

B. Equilibrium constant determination.

### **RECOMMENDED BOOKS:**

1. Advanced Experimental Physical Chemistry by Ayodhya Sing.
2. Experimental Physical Chemistry by Daniel
3. Experimental Physical Chemistry by G.Peter Matthews.
4. Experiments in Physical Chemistry by Shoemake.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-406</b>	<b>Inorganic Chemistry (Sp. Theory-I)</b>	<b>4</b>	<b>VII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### **SYLLABUS OUTLINE:**

**1. Stereochemistry and bonding in main group compounds:**

Introduction, directed valence theory, three center bond model, correlation diagram approach, some qualitative failures of the simple theories criticism and comparison of simple models,  $d\pi$ - $P\pi$  bonds.

**2. Periodicity:**

First and second row anomalies. The use of d-orbitals by non-metals, reactivity and d-orbital participation. The use of p-orbitals in  $\pi$ -bonding, periodic anomalies of the non-metals and post-transition metals.

**3. Organic Reagents used in Inorganic Analysis:**

Classification of organic reagents, their selectivity and specificity, methods of preparation of specific compounds and their studies with UV, Visible and IR. Typical reagents used in complexometric titrations involving the use of EDTA. Chelates, classification, stability, preparation and properties. Role of organic Reagents in different analytical techniques.

### **RECOMMENDED BOOKS:**

1. Organotransition metal Chemistry by Akin Yamamoto, 1996, A. Wiley Interscience Publication London.
2. Hand Book of Organic reagents in Inorganic Analysis by ZAVIX Holzbecher and other 1976 Ellis Hurwod Limited, London.
3. Structural Inorganic Chemistry by Wells, A.F. 1975, Charenden Press, London.
4. Stereochemistry and bonding in Inorganic Chemistry by by J.E. Ferguson 1974, Prentice Hall, New Jersey.
5. J H Huheey, Inorganic Chemistry - Principles, structure and reactivity, Harper and Row Publisher, Inc. New York (2008)
6. Cullen Dolphin and James, Biological aspects of Inorganic Chemistry, 2005.
7. Williams, An Introduction to Bioinorganic Chemistry, 2003.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-407</b>	<b>Inorganic Chemistry (Sp. Theory-II)</b>	<b>4</b>	<b>VII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### **SYLLABUS OUTLINE:**

#### **1. Chemistry of Oxides:**

Physical states and structures of oxides of the elements, covalent oxides, periodic trends in structure and physical state, acidity, solubility, practical uses, and environmental chemistry of volatile oxides, close packed anions, metal oxides, electrical conductivity of solid ionic compounds, spinels, perovskites, high temperature superconductors, magnetic properties in mixed metal oxides.

#### **2. Reactions in aqueous and non-aqueous solvents:**

Classification of solvents, types of reactions, the dielectric constant, solubilities, electrode potential and electromotive forces. Reactions in water and molten salts. Reactions in non-aqueous solvents, i.e. ammonia, sulphur dioxide, bromine trifluoride and hydrofluoric acid.

#### **3. Radioactivity:**

Natural radioactivity, Artificial radioactivity, types of radioactive rays, Saddy-Fajans and Russel group displacement law, Half life period of a radioactive substance, Disintegration constant K, Average life period, Radioactive equilibrium, Law of successive disintegration, Activity of a radioactive substance, Transmutation of elements, Artificial transmutation reactions induced by different bombarding projectiles, Applications of artificial transmutation reactions, Natural and artificial radioactive series.

### **RECOMMENDED BOOKS:**

1. J H Huheey, Inorganic Chemistry - Principles, structure and reactivity, Harper and Row Publisher, Inc. New York (2008)
2. Cullen Dolphin and James, Biological aspects of Inorganic Chemistry, 2005.
3. Williams, An Introduction to Bioinorganic Chemistry, 2003
4. Organotransition metal Chemistry by Akin Yamamoto, 1996, A. Wiley Interscience Publication London.
5. Hand Book of Organic reagents in Inorganic Analysis by ZAVIX Holzbecher and other 1976 Ellis Hurwod Limited, London.
6. Structural Inorganic Chemistry by Wells, A.F. 1975, Charenden Press, London.
7. Stereochemistry and bonding in Inorganic Chemistry by J.E. Ferguson 1974, Prentice Hall, New Jersey.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-408</b>	<b>Inorganic Chemistry (Sp. Practical)</b>	<b>2</b>	<b>VII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### SYLLABUS OUTLINE:

**1. Use of organic reagents for the estimation of various metal ions;  
(At least any four of the following):**

- (a) 8-Hydroxyquinoline ( $Al^{3+}$ ,  $Ti^{3+}$ ,  $Fe^{3+}$ )
- (b) Pyrogallol ( $Bi^{3+}$ )
- (c) Nitron ( $NO_3^{-}$ )
- (d) Salicyladoxime ( $Ni^{2+}$  in presence of  $Cu^{2+}$ )
- (e) Anthranilic acid ( $Cd^{2+}$ ,  $Zn^{2+}$ ,  $Co^{2+}$ )

**2. Instrumental methods of analysis:**

**(a) Conductometry:**

- i). Titration of HCl and acetic acid with a strong base.
- ii). Precipitation titration of  $AgNO_3$
- iii). Determination of  $K_a$  for acetic acid.

**(b) Colorimetry:**

- i). Micro determination of chromium by diphenyl Carbazide.
- ii). Determination of iron by 1, 10 Phenanthroline.
- iii). Determination of nickel by rubeanic acid

### RECOMMENDED BOOKS:

1. Hand Book of Organic reagents in Inorganic Analysis by ZAVIX Holzbecher and other 1976 Ellis Hurwod Limited, London.
2. J. Bassett, R. C. Denny, G. H. Jeffery and J. Mendham, Vogel's Text Book of qualitative Inorganic Analysis, the English Language Book Society and Longman, New York, (2008)
3. Quantitative Analysis Chemistry, James S. Pritz, George H. Sehenk, 2001 Alby and Becon Inc. London.
4. Theory and practice of chromatography by Prof. Dr. Javed Iqbal (2002).
5. Instrumental analysis by Gary D. Christian and James E.O., Reilly, 2007, Allyn and bacon Inc., London.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-425	Inorganic Chemistry (Sp. Theory-I)	4	VIII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### 1. Advances in Atomic Spectroscopy and Radio Chemistry:

Various forms of interaction of electromagnetic radiation with matter, absorption and emission spectra. Flame photometer. Atomic absorption spectrophotometer, Inductively coupled plasma emission, Applications in clinical chemistry, industry, geochemistry, soil sciences and environmental studies.

#### 2. Some Thermodynamic aspects of Inorganic Chemistry:

Thermodynamic and kinetic stability, Interpretation of stability, Role of thermodynamics in interpretative chemistry, The lattice energy as a criterion of bond type, Quantitative uses of the lattice energy, The Kapustunskii equations, The stabilization of high oxidation states by fluorine and oxygen, The stabilization of low oxidation states by large anions, Halogen exchange reaction, The stability of halides containing protonated bases.

#### 3. Polymeric Inorganic Compounds (Chains, Rings and Cages):

- (a) Chains: Catenatin, Homocatenotion, Heterocatenation, Silicones, Silicates, Zeolites, talc, mica, clay.
- (b) Rings: (i) Heterocyclic systems of borazines, Phosphazenes, S-N rings.  
(ii) Homocyclic system of sulphur and selenium.
- (c) Cages compounds of phosphorus, and boron
- (d) Inorganic Polymers as Conductors.

### RECOMMENDED BOOKS:

1. Structural Inorganic Chemistry by Wells, A.F. 1975, Charenden Press, London.
2. Stereochemistry and bonding in Inorganic Chemistry by by J.E. Ferguson 1974, Prentice Hall, New Jersey.
3. J H Huheey, Inorganic Chemisry - Principles, structure and reactivity, Harper and Row Publisher, Inc. New York (2008)
4. Analytical chemistry, G.D Christian 2005

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-426</b>	<b>Inorganic Chemistry (Sp. Theory-II)</b>	<b>4</b>	<b>VIII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### SYLLABUS OUTLINE:

#### **1. Inorganic Chemistry in Biological systems:**

Energy sources for life, metalloporphyrins. Photosynthesis and respiration. Nitrogen fixation, the biochemistry of Iron essential and trace elements in biological systems biochemistry of the nonmetals medicinal chemistry organometallic in bio Inorganic Chemistry.

#### **2. Kinetics and mechanisms of Reaction of Coordination Compounds:**

Introduction of reaction rate law and mechanism of stationary state approximation. Type of reactions, nucleophilic displacements, effective collisions. Dis-placement in square planar complexes, trans-effect, replacement in octahedral complexes, inert and labile complexes, (VBT, CFT explanation), Inner and outer sphere exchange reactions.

#### **3. Organo Metallic Compounds:**

**(Synthesis, Structure, Bonding & Reaction Pathways)**

Nature of carbon-metal bond, classification, synthesis and properties of organometallic compounds ( $\sigma$ -bonded olefin,  $\pi$ -allylic,  $\pi$ -cyclopentadienyl,  $\pi$ -organometallic compounds) and characterization of organometallic compounds with the help of IR, NMR, mass spectrometry etc. Experimental techniques in Organometallic chemistry oxidative-addition, reductive elimination, insertion and de-insertion reactions, fluxional behaviour. Applications of organometallic compounds.

### RECOMMENDED BOOKS:

1. Cullen Dolphin and James, Biological aspects of Inorganic Chemistry, 2005
2. Williams, An Introduction to Bioinorganic Chemistry, 2003
3. Organotransition metal Chemistry by Akin Yamamoto, 1996, A. Wiley Interscience Publication London.
4. Hand Book of Organic reagents in Inorganic Analysis by ZAVIX Holzbecher and other 1976 Ellis Hurwod Limited, London.
5. Structural Inorganic Chemistry by Wells, A.F. 1975, Charenden Press, London.
6. Stereochemistry and bonding in Inorganic Chemistry by J.E. Ferguson 1974, Prentice Hall, New Jersey.
7. J H Huheey, Inorganic Chemistry - Principles, structure and reactivity, Harper
8. and Row Publisher, Inc. New York (2008)

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-427	Inorganic Chemistry (Sp. Practical)	2	VIII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### 1. Chromatographic Techniques:

- (a) Column and thin layer techniques for the qualitative analysis of inorganic compounds.
- (b) Applications of solvent extraction and ion exchangers technique.

#### 2. Instrumental Methods of Analysis:

- (a) Potentiometry:
  - i). Determination of  $K_1$ ,  $K_2$  and  $K_3$  for  $H_3PO_4$
  - ii). Determination of Cobalt (II)
  - iii). Determination of iron (II)
  - iv). Determination of chloride in presence of iodide and evaluation of  $K_{sp}$  for AgI and AgCl.
- (b) Thermal analysis  
Decomposition studies of the following substances:
  - i). Copper sulphate penta hydrate
  - ii). Calcium oxalate monohydrate
- (c) Atomic absorption spectroscopy  
Estimation of following:  
 $Mg^{2+}$ ,  $Zn^{2+}$ ,  $Al^{3+}$

### RECOMMENDED BOOKS:

1. J. Bassett, R. C. Denny, G. H. Mendham, Vogel's Text Book of qualitative Inorganic Analysis, the English Language Book Society and Longman, New York, (2008).
2. Quantitative Analysis Chemistry, James S. Pritz, George H. Sehenk, 2001 Alby and Becon Inc. London.
3. Theory and practice of chromatography by Prof. Dr. Javed Iqbal (2002).
4. Instrumental analysis by Gary D. Christian and James E.O., Reilly, 2007, Allyn and bacon Inc., London.
5. Hand Book of Organic reagents in Inorganic Analysis by ZAVIX Holzbecher and other 1976 Ellis Hurwod Limited, London.
6. Experimental Inorganic Chemistry - W. G. Palmer, 2005.
7. The analysis of minerals and ores of the rarer elements – W. R. Schoeller, and A. R. Powell, Charles, Griffin and Company Limited, 2004.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-409	Organic Chemistry (Sp. Theory-I)	4	VII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

**1. Determination of Reaction Mechanism:**

Determination of reaction mechanism, kinetics, stereochemical, intermediate formation, spectroscopic and isotopic labeling methods

**2. Aliphatic Nucleophilic substitution:**

Mechanism of SN1, SN2, SNi, SN1', SN2' and SNi' reactions, kinetics, stereochemical and other evidence; effects of other substrate structure, attacking nucleophile, leaving group and solvent effect, and neighbouring group participation.

**3. Elimination Reactions:**

Mechanism of E1, E2, and E1cb elimination reactions; kinetics and stereochemical studies; applications of thermodynamically and kinetically controlled reactions (Saytzeff and Hoffmann reactions), Effects of substrates, solvent, base, leaving group and temperature on kinetics, competition between elimination and substitution reactions; pyrolytic elimination reaction mechanism and synthetic applications.

### RECOMMENDED BOOKS:

1. Organic Chemistry, Volume I (6<sup>th</sup> ed.) & II (5<sup>th</sup> d.) by I.L. Finar, Pearson Education (singapore) Pte Ltd, 2008.
2. March's Advanced Organic Chemistry: Reactions, Mechanisms, and Structure, 6th ed. by Michael B. Smith, Jerry March, Wiley, 2007.
3. Organic Chemistry, 5th ed.; by S. H. Pine, McGraw Hill: New York, 1987.
4. Organic Chemistry 6th ed. by Francis A. Carey, McGraw Hill, 2005.
5. Organic Chemistry 6<sup>th</sup> ed, by R. T. Morrison, R. N. Boyd, and R. K. Boyd, Benjamin Cummings, 1992,.
6. Modern Synthetic Reactions 2nd ed. by H.O.House, W.A. Benjamin Inc., Menlo Park, CA
7. Principles in Organic Synthesis by R.O.C Norman & J. M. Coxon, Chapman and Hall, 1993.
8. Organic Chemistry by Jonathan Clayden, Nick Greeves, Stuart Warren, Oxford University Press 2000.



## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM- 410	Organic Chemistry (Sp. Theory-II)	4	VII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### 1. Aromatic Substitution reactions:

##### (a) Electrophilic substitution:

General mechanism (kinetic, isotopic and spectroscopic evidences), nitration, sulfonation, halogenation, Friedel-Crafts alkylation and formylation, structure and reactivity, orientation; polysubstitution reactions of aromatic compounds

##### (b) Nucleophilic Substitution reactions:

$S_N1$ ,  $S_N2$  (addition and elimination), and Benzyne mechanism

#### 2. Molecular Rearrangements:

Classification of molecular rearrangements: mechanism of intramolecular 1,2-shifts involving migration of a group from carbon to carbon, carbon to nitrogen, and carbon to oxygen, mechanism and synthetic applications of Wagner-Meerwein, Pinacol-pinacolone, benzidine, benzyl, benzylic acid, Favorski, Wolff, Beckmann, Hoffmann, Curtius, Lossen and Schmidt; Baeyer-Villiger, Dakin and Fries rearrangements.

#### 3. Heterocyclic Chemistry:

Five and six membered heterocycles with one and several identical hetero-atoms, Five and six membered heterocycles with two different hetero-atoms.

### RECOMMENDED BOOKS:

1. Organic Chemistry, Volume I (6<sup>th</sup> ed.) & II (5<sup>th</sup> ed.) by I.L. Finar, Pearson Education (Singapore) Pte Ltd, 2008.
2. March's Advanced Organic Chemistry: Reactions, Mechanisms, and Structure, 6th ed. by Michael B. Smith, Jerry March, Wiley, 2007.
3. Organic Chemistry, 5th ed.; by S. H. Pine, McGraw Hill: New York, 1987.
4. Organic Chemistry 6th ed. by Francis A. Carey, McGraw Hill, 2005.
5. Organic Chemistry 6<sup>th</sup> ed, by R. T. Morrison, R. N. Boyd, and R. K. Boyd, Benjamin Cummings, 1992,.
6. Heterocyclic Chemistry, 4<sup>th</sup> ed. by J. A. Joule, K. Mills, Blackwell Publishing, 2000.
7. Heterocyclic Chemistry, 3rd ed. by T.L. Gilchrist, Longman, 1997.
8. Principles in Organic Synthesis by R.O.C Norman & J. M. Coxon, Chapman and Hall, 1993.
9. Organic Chemistry by Jonathan Clayden, Nick Greeves, Stuart Warren, Oxford University Press 2000.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-411</b>	<b>Organic Chemistry (Sp. Practical)</b>	<b>2</b>	<b>VII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### **SYLLABUS OUTLINE:**

**1. Qualitative analysis:**

Three component organic mixture analysis (separation and identification of the unknown components)

**2. Chromatography and Spectroscopy:**

Separation of mixtures by chromatography and identification by spectroscopy

### **RECOMMENDED BOOKS:**

1. Practical Organic Chemistry by F. G. Mann and B. C. Saunders, 1978, Longman, UK, 1978.
2. Vogel's Textbook of Practical Organic Chemistry (5th ed.) by A.I.Vogel, A.R.Tatchell, B.S. Furnis, A.J. Hannaford, P.W.G. Smith, Longman UK, 1989.
3. The Systematic Identification of Organic Compounds,(8th ed.) by Ralph L. Shriner et al., Wiley, 2003.
4. Advanced Practical Organic Chemistry, by J. Leonard, B. Lygo, G. Procter, CRC, 1994.
5. Advanced Practical Organic Chemistry (2nd ed.) by N. K. Vishnoi, Vikas Publishing House Pvt Ltd, India, 1996.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM- 428	Organic Chemistry (Sp. Theory-I)	4	VIII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### 1. Nuclear Magnetic Resonance Spectroscopy:

Nuclear magnetic resonance: Basic principles, theory, spin flipping, nuclear precession and absorption of electromagnetic radiation, spin relaxation, basic introduction of 1-D ( $^1\text{H}$  and  $^{13}\text{C}$ ) NMR spectroscopy, chemical shifts and integration curve, instrumentation, spin-spin splitting and coupling constants. Structure elucidation of small molecules. Introduction of 2-D NMR spectroscopy.

#### 2. Mass Spectroscopy:

Introduction; types, Isotopic abundance, molecular and metastable ions; fragmentation pattern, applications of mass spectroscopy in different classes of organic chemistry, interpretation of mass spectra of small organic molecules.

#### 3. Natural Products:

Introduction, classification, isolation, biosynthesis and general methods for the structure determination of alkaloids (piperine, Nicotene, Cocaine, Morphine, Quinine), steroids (cholesterol, progesterone, *estrogens*, androgens, glucocorticoids mineralocorticoids) and terpenoids (Triterpenes,  $\alpha$ -amyrin,  $\beta$ -amyrin, Ursolic acid, Oleanolic acid).

### RECOMMENDED BOOKS:

1. Organic Chemistry, Volume II by I.L. Finar; 5th ed. Longman scientific, 1975.
2. Spectroscopic Methods in Organic Chemistry 6th ed. by D. Williams and I. Fleming. Wiley-VCH, 1991.
3. Spectrometric identification of Organic Compounds 6th ed. by R. M. Silverstein and F. X. Webster, Wiley, 2007.
4. Organic Spectroscopy and Chromatography by M Younas, ILMI, Pakistan, 2007.
5. Spectroscopy by Pavia, Lampman, Kriz, 2nd ed., Harcourt Brace College Publishers, 1996.
6. Biosynthesis of Natural Products, Paolo Manito, John Wiley & Sons, 1980.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-429</b>	<b>Organic Chemistry (Sp. Theory-II)</b>	<b>4</b>	<b>VIII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### SYLLABUS OUTLINE:

#### **1. Reactive Intermediates:**

Carbenes, nitrenes, and Arynes, structure and evidence for formation, general reactions and synthetic applications

#### **2. Organic Synthesis:**

An outline of the recent developments in organic syntheses involving boron, phosphorus, silicon reagents, reaction conditions and methods; phase transfer catalysis involving quaternary ammonium, phosphonium salts and crown ethers, solid phase synthesis, Introduction to protective groups, protection of hydroxyl, amino, carbonyl and carboxylic groups, reactions involving the introduction and removal of some common protective groups and synthetic applications. Introduction to disconnection approach.

#### **3. Pericyclic reactions:**

Introduction, Woodward-Hoffmann rules and molecular orbital theory; cycloaddition, electrocyclic and sigmatropic rearrangement and group transfer reactions.

### RECOMMENDED BOOKS:

1. Organic Chemistry, Volume I (6th ed.) & II (5th ed.) by I.L. Finar, Pearson Education (Singapore) Pte Ltd, 2008.
2. March's Advanced Organic Chemistry: Reactions, Mechanisms, and Structure, 6th ed. by Michael B. Smith, Jerry March, Wiley, 2007.
3. Organic Chemistry, 5th ed.; by S. H. Pine, McGraw Hill: New York, 1987.
4. Organic Chemistry 6th ed. by Francis A. Carey, McGraw Hill, 2005.
5. Organic Chemistry 6th ed, by R. T. Morrison, R. N. Boyd, and R. K. Boyd, Benjamin Cummings, 1992,.
6. Phase-transfer catalysis: fundamentals, applications, and industrial perspectives by C. M. Starks, C.L. Liotta and M. Halpern, Chapman & Hall, 1994.
7. Principles in Organic Synthesis by R.O.C Norman & J. M. Coxon, Chapman and Hall, 1993.
8. Organic Chemistry by Jonathan Clayden, Nick Greeves, Stuart Warren, Oxford University Press 2000.
9. Electrocyclic Reactions by F. L. Ansari, R. Qureshi, M. L. Qureshi, 1999, Wiley-VCH.
10. Reactive Intermediates in Organic chemistry, N. S. Isaac, John Wiley and Sons, 1974.
11. Work book for Organic Synthesis, The disconnection approach, Stuart Warren, John Wiley and Sons, 1994,.
12. Organic Synthesis, The disconnection approach, Stuart Warren, 1993, John Wiley and Sons 1993.

13. Designing Organic Synthesis, A Programmed Introduction to synthon approach, S. Warren, John Willey and Son, 1992.
14. Guide book to Organic Syntheses, R. K. Mackie, D. M. Smith, Longman Group Limited, 1982.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-430	Organic Chemistry (Sp. Practical)	2	VIII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### Organic Synthesis:

- (a) Multi-step preparation and spectroscopic characterization: *p*-nitroaniline from aniline; *p*-bromotoluene from *p*-toluidine, *o*-Bromotoluene from *o*-toluidine
- (b) Preparation, separation and identification of regio-isomers: *o*-nitrotoluene and *p*-nitrotoluene from toluene; *o*-nitrosophenol and *p*-nitrosophenol from phenol

### RECOMMENDED BOOKS:

1. Practical Organic Chemistry by F. G. Mann and B. C. Saunders, 1978, Longman, UK, 1978.
2. Vogel's Textbook of Practical Organic Chemistry (5th ed.) by A.I. Vogel, A.R. Tatchell, B.S. Furnis, A.J. Hannaford, P.W.G. Smith, Longman UK, 1989.
3. Advanced Practical Organic Chemistry, by J. Leonard, B. Lygo, G. Procter, CRC. 1994.
4. Advanced Practical Organic Chemistry (2nd ed.) by N. K. Vishnoi, Vikas Publishing House Pvt Ltd. India, 1996.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-412	Analytical Chemistry (Sp. Theory-I)	4	VII
Year	Discipline		
4	Chemistry		

### **SYLLABUS OUTLINE:**

#### **Gas Liquid Chromatography / Gas Solid Chromatography:**

Gas Chromatographs, Derivative Formation, Gas Chromatographic Columns, Liquid Phases and Column Selection, Detectors for Gas Chromatography, Optimization of Experimental Condition, Gas-Solid Chromatography, Interfacing Gas Chromatography with Mass Spectrometry, Interfacing Gas Chromatography with Infrared Spectrometry,

#### **High Performance Liquid Chromatography:**

Optimization of Column Performance, Gradient Elution and Related Procedures, Derivation, HPLC Instrumentation, Mobile-Phase Delivery System, Sample Introduction, Separation Columns, Detectors, Interfacing HPLC with Mass Spectrometry

#### **Potentiometry:**

Nernst equation; Electrode Potentials; different reference electrodes including glass and calomel electrode; working of a potentiometer and its applications including pH measurements and potentiometric titrations; ion-selective electrode systems; ion-exchange membrane electrode; gas-sensing electrode; solid-state membrane electrode and bio membrane electrode.

#### **Thermo gravimetric Analysis / Differential Thermal Analysis:**

General Principle of thermal, instrumentation, types of measurements; TGA (thermogravimetric analysis), DTA (differential thermal analysis), DSC (differential scanning calorimetry), TT (thermometric titrations) and EGD (evolved gas detection), Principles, instrumentation and applications of these techniques.

### **RECOMMENDED BOOKS:**

1. Electro Analytical Chemistry by J.J. Longane, Inter Science Publisher Inc. N.Y. London.
2. Vogels, text book of Quantitative chemical analysis by J. Mendham, R.C. Denney, J.D. Barnes, M.J. K. Thomas, Pearson Education Ltd.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-413</b>	<b>Analytical Chemistry (Sp. Theory-II)</b>	<b>4</b>	<b>VII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### **SYLLABUS OUTLINE:**

#### **FTIR / Raman Spectroscopy:**

Origin of Infra Red Spectra; Different vibrational modes, Normal coordinate and normal vibrations, Symmetry of normal vibrations and selection rule, Raman Spectroscopy, Metal-isotope spectroscopy, Vibrational Spectra in gaseous phase and inert gas matrices; Comparison of raman with Infra Red spectroscopy; Applications for qualitative and quantitative chemical analysis; Instrumentation details and their function.

#### **UV / Vis Spectroscopy:**

The Nature of Electromagnetic Radiation, The Electromagnetic Spectrum, Atomic Energy Levels, molecular Electronic Energy Levels, Vibrational Energy Levels, Raman Effect, Lasers,

Radiation Sources, Wavelength Selection, Cells and Sampling Devices, Detectors, Readout Modules, Instruments for Absorption Photometry

#### **Atomic / Mole Fluorescence:**

Atomic Fluorescence Spectroscopy; Instrumentation, applications and limitations of these techniques

Plasma Source

Inductively coupled plasma sources, special detection systems and read out devices used for ICPEs; multielement analysis with plasma devices.

### **RECOMMENDED BOOKS:**

1. Chemical Application of Spectroscopy by West, Inter Science Publisher Inc. N.Y. London.
2. Kinetics in Analytical Chemistry by H.B. Mark Jr. & G.A. Rechnitz, Interscience N.Y. (1968).
3. Analytical Chemistry by Gary D. Christian, John Wiley and Sons (1977).
4. Automated Chemical Analysis by J.K. Forman Stockwell, John Wiley and Sons, N.Y. (1975).
5. Advances in Infrared Group Frequencies by L.J. Bellacy, Mathuen & Col. Amsterdam (1968).



## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-414</b>	<b>Analytical Chemistry (Sp. Practical)</b>	<b>2</b>	<b>VII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### **SYLLABUS OUTLINE:**

#### **Molecular Spectrophotometry:**

- Determination of Iron (II) using 1, 10-phenanthroline method.  
Determination of Iron (III) using thiocyanate method.  
Determination of chromium by diphenylcarbazide method.  
Determination of Fe, Pb, Cd, Zn and Cu in soil samples by AAS technique.  
Simple acid base titrations using potentiometer.  
Determination of "F" in water by using ion selective electrodes.  
Preparation of standard calibration graphs of Pb, Cd, Zn and Fe by AAS.  
Determination of Ni by DMG method spectrophotometrically.  
Determination of mixtures of complexes of Iron with Thiocyanide and 1, 10, phenanthroline.  
Determination of  $\lambda_{max}$  of Cr complex with diphenyl carbazide.  
Infrared Determination of a Mixture of Xylene  
Spectrophotometric determination of Lead or Leaves using Solvent Extraction.

#### **RECOMMENDED BOOKS:**

1. Vogel's, a text book of quantitative inorganic analysis by J. Bassett. The English language book Society and Longman.
2. Atomic and molecular spectroscopy. Basic concepts and practical application by Sune Svanberg, Springerlink Berlin Heidelberg.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-431	Analytical Chemistry (Sp. Theory-I)	4	VIII
Year	Discipline		
4	Chemistry		

### **SYLLABUS OUTLINE:**

#### **Voltrametry:**

Principle and applications of anodic stripping voltametry

#### **Polarography:**

Introduction and principle of polarography, basic instrumentation, working and advantages of DME (dropping mercury electrode); limiting and residual current; half-wave potential; qualitative and quantitative aspects of polarographical analysis

#### **Amperometry:**

#### **Conductometry:**

Conductance in Solutions; Specific conductance; molar conductance; factors upon which the conductance of solution depends; Measurement of conductance; cell constant; Analytical applications of conductance measurement.

#### **Arc / Spark / Glow, Automation:**

### **RECOMMENDED BOOKS:**

1. Vogels, s text book of quantitative inorganic analysis by J. Bassett. The English language book Society and Longman.
2. Vogels, text book of Quantitative chemical analysis by J.mendham, RC Denny, JDBarnes, MJ KTHomas, Pearson education Ltd.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-432	Analytical Chemistry (Sp. Theory-II)	4	VIII
Year	Discipline		
4	Chemistry		

### **SYLLABUS OUTLINE:**

#### **Laser Spectroscopy:**

Principle of laser operation; Stimulated emission Population inversion, Single level and multi-level laser systems, Properties of laser light and its general and analytical applications; ruby laser, nitrogen laser, dye laser, Use of laser radiation in absorption and fluorescence spectroscopic methods.

#### **Nuclear Magnetic Resonance Spectroscopy:**

Basic principles; properties of nuclei, Chemical shifts; Spin-Spin coupling; Pulsed Fourier Transform NMR Spectrometry; Identification of structural features; Use of NMR imaging in medicine; Analytical applications of NMR spectroscopy.

#### **Mass Spectrometry:**

Principle of mass spectrometry; Intel system, ionization, acceleration, Drift Chamber, Detection systems; Advancements in equipment in equipment; Analytical uses of mass spectrometry, Quadrupole mass spectrometry; Interpretation of mass spectra. Correlation of mass spectra with Molecular structure.

### **RECOMMENDED BOOKS:**

1. New Instrumental Methods in Electro Chemistry by Faul-Delabay, Inter Science Publisher, London, N.Y.
2. Instrumental Methods of Analysis by Hobert H. Willart, Lyle L. Merrit, D. Van Nosrant Company Inc. N.Y. London.
3. Principles of Polarography by J. Herosky & J. Kuta, Academic Press N.Y. (1968).
4. Laser spectroscopy by Wolfgang Demtroder, springerlink.
5. Analytical chemistry by Kellner, J.M. Mermet, Wiley-VCH Verlag GmbH & Co. KGaA.
6. A text book of analytical chemistry by Y-Anjaneyulu, K-chamdare khar, ValiManickam, Pharma book syndicate.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-433</b>	<b>Analytical Chemistry (Practical)</b>	<b>2</b>	<b>VIII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### **SYLLABUS OUTLINE:**

#### **Conductometry:**

Determine the amount of HCl conductometrically by using strong base NaOH.

Determine the amount of base NH<sub>4</sub>OH conductometrically by using strong acid.

Determine the amount of NH<sub>4</sub>OH by using weak acid CH<sub>3</sub>COOH conductometrically.

Determine the amount of NaOH conductometrically by using weak acid CH<sub>3</sub>COOH.

#### **Potentiometry:**

Determine the amount of HCl by using strong base (NaOH) potentiometrically.

Determine the amount of HCl by using weak base (NH<sub>4</sub>OH) potentiometrically.

Determine the amount of CH<sub>3</sub>COOH by using strong base (naoh).

Determine the amount of HCl & CH<sub>3</sub>COOH conductometrically by using strong base NaOH.

### **RECOMMENDED BOOKS:**

1. Vogels, text book of Quantitative chemical analysis by J. Mendham, R. C. Denny, J. D. Barnes, M. J. Thomas, Pearson Education Ltd.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-415	Applied Chemistry (Sp. Theory-I)	4	VII
Year	Discipline		
4	Chemistry		

### **SYLLABUS OUTLINE:**

#### **1. Processing of Crude Oil:**

Brief description of origin of petroleum, Migration, reservoir, exploration and drilling of crude petroleum; Various unit processes involved petroleum processing like surface operation, fractional distillation; refining, cracking; reforming, isomerization; Polymerization, alkylation and finishing processes

#### **2. Paper and Pulp Industries:**

History and back ground, survey of raw materials, production of pulp by soda process, sulphite process and Kraft (sulphate) process, Manufacture of paper environmental aspects of paper industry.

#### **3. Fertilizer Industries (Ammonia, Urea and other Fertilizers):**

##### **(a) Ammonia:**

Raw materials, various sources of hydrogen and nitrogen, manufacture of ammonia (Haber's process), its use as fertilizer and other applications.

##### **(b) Urea:**

Raw materials, manufacture of urea, assimilation in soil.

##### **(c) Calcium Fertilizers:**

Calcium ammonium nitrate, calcium cyanamide, calcium super phosphate and triple super phosphate.

##### **(d) Potash Fertilizers:**

Manufacture and use of potash fertilizers.

### **RECOMMENDED BOOKS:**

1. Pulp and Paper Technology, Testing and Applications, K.P. Rao (2003), CBS Publishers.
2. Chemistry of Pulp and Paper making, Edwin Sutermeister, Ed-3<sup>rd</sup> (1946)
3. Fertilizers and Soil Fertility, U.S.Jones, Reston Publishing Co. Virginia, 1979.
4. Petroleum Refining Technology, Ram Parsad (2002).
5. Industrial chemistry, B. K. Sharma, Krishna Prakashan Media (P) Ltd., Ed-15 (2006).
6. Shereve's Chemical Process Industries, 5th Ed. 1975, by G.T. Austin, McGraw Hill Book Co. New York.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-416	Applied Chemistry (Sp. Theory-II)	4	VII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### 1. Steel, Metallurgical Products and Electroplating:

##### (a) **Steel and Metallurgical Products:**

Manufacture of cast iron and steel, wrought iron, formation of alloys, heat treatment of steel, classification of steel, passivity, different theories of rusting of iron and its prevention.

##### (b) **Electroplating:**

Principle of electroplating, purpose of electroplating, different processes involved in electroplating, chrome plating, nickel plating and electroplating of plastics.

#### 2. Textile Dyeing:

Color and chemical constitution, Important classes of chromogens, Classification and nomenclature of dyes, manufacturing of dye intermediates and dyes, Selection of dyes for wool, cellulosic and synthetic fibers, Theory of Coloration, Coloration of wool, cellulosic and synthetic fibers.

#### 3. Analytical Techniques and on-line analysis in Industry:

##### (a) **Spectroscopy:**

Use of different spectroscopic techniques like FES, AAS and Spectrophotometry for the quality control of raw materials, intermediates and final products in various industries.

##### (b) **Chromatography:**

Use of Thin layer chromatography, gas chromatography and HPLC in pharmaceutical and other industries.

##### (c) **On-Line Analysis and Automation:**

Significance of on-line analysis and automation of analytical techniques in industry; Classification of techniques w.r.t. automation, Use of microprocessor in conjugation with automation and on-line analysis, different types of automatic analyzers.

### RECOMMENDED BOOKS:

1. Chemistry of iron and Steel Manufacture, C.Bodsworth, Longman Press, London, 1963.
2. Instrumental method of Analysis, Willar Merritt Dean Settle, Seventh Edition (1986)
3. Graham's Electroplating Engineering Hand Book, Ed. L.J. Durney, CBS Publishers and Distributors, New Delhi. (1997).
4. Nickel and Chromium plating, J.K.Dennis & T.E.Such, Newness Butterworth, London (1972).
5. T. B. of Quantitative Inorganic Analysis, Vogel's Ed-4<sup>th</sup>, Longman Group Limited (1978).
6. Instrumental Analysis, Gary D. Christain, 1978, Introduction to Instrumental Analysis by Braun, McGraw-Hill Book company, 1987.

7. Dyes and Dyeing, C.E. Pellow, Abhishek Publishers, 1998.
8. Textile Dyes and Pigments, H. Panda, NIIR Publishers.
9. The Chemistry of Synthetic Dyes and Pigments, H. A. Lubs, Reinhold Publishing Corporation, 1955.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-417	Applied Chemistry (Sp. Practical)	2	VII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### 1. Oil & Fats

Acid value; Ester Value; Saponification value and Iodine value of different vegetable oils

#### 2. Water Pollution

- (a) Dissolved Oxygen
- (b) Biological Oxygen Demand
- (c) Chemical Oxygen Demand

#### 3. Industrial Analysis

Steel Analysis involving volumetry, spectrophotometry and solvent extraction.

Analysis of dolomite; haemetite; chromite and bauxite ores involving separation and analytical techniques.

#### 4. Textile

Dyeing of clothes in different shades using Acid dyes

### RECOMMENDED BOOKS:

1. Metallurgical Analysis, Lord (1893).
2. Applied Chemistry Theory and Practice, O.P. Vermani & A.K. Narula, Wiley Eastern Limited (1989).
3. T. B. of Quantitative Inorganic Analysis, Vogel's Ed-4<sup>th</sup>, Longman Group Limited (1978).
4. Practical Statistics for the Analytical Scientist, A Bench Guide, RSC Publishing LGC Ltd 2009.
5. Dyes and Dyeing, C. E. Pellew, Abhishek Publishers (1998).



## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-434	Applied Chemistry (Sp. Theory-I)	4	VIII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### 1. Leather Tanning Industries:

Introduction, important steps in leather manufacturing, theory of leather tanning, waste disposal and pollution aspects involved in tanning industries.

#### 2. (a) Oils and Fats:

Classification of oils and fats, vegetable oils, essential oils, various methods of extraction of oils, refining and hydrogenation of oils, Industrial applications of oils in resins, surfactants, lubricants and paints.

#### (b) Petrochemicals:

Chemistry and importance of the following petrochemicals: Acetylene; ethylene; propylene, benzene, toluene, xylene and naphthalene. Oxidation halogenation and nitration of petrochemicals of industrial significance.

#### 3. Textile Fibers:

Classification of synthetic fibers, chemistry and manufacturing of viscose rayon, true synthetic fibers including nylons and polyester fibers. Finishing processes for 100% cotton fabrics such as singeing, desizing, scouring, mercerizing and bleaching.

### RECOMMENDED BOOKS:

1. Food Oils and Fats, H.Lawson, CBS Publishers and Distributors, New Delhi.(1997).
2. Industrial chemistry, B. K. Sharma, Krishna Prakashan Media (P) Ltd., Ed-15 (2006).
3. Leather in life art and industry, John W. Waterer, 1927. Faber and Faber Ltd.
4. Fibre to fabric, 4<sup>th</sup> Ed, Potter & Corban, McGraw Hill book Company, 1959.
5. A Text on Petrochemicals, Bhaskararsao, 2002.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-435	Applied Chemistry (Sp. Theory-II)	4	VIII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

1. **Coal Chemicals fuel Gases:**

The destructive distillation of coal, coking of coal distillation of coal tar; Liquid Fuels: Hydrogenolysis Natural gas; Coal Gas: Water Gas; Liquefied Petroleum Gases.

2. **Sugar Industry and Related Chemicals:**

Extraction of juice from sugar cane, purification of juice, clarification, concentration, refining and crystallization. Beet sugar. Glucose and hydrogenated glucose syrups.

3. **Polymers:**

General classification of polymers, characteristics and significance of polymers, various mechanisms of polymerization process, Polymer processing like extrusion, injection, modeling and blow molding of plastics.

Brief description and uses of the following polymers:

Polyethylene, polystyrene, epoxy resins, polyethylene tetraphthalate.

### RECOMMENDED BOOKS:

1. Shereve's Chemical Process Industries, 5th Ed.1975, G.T.Austin, McGraw Hill Book Co. New York.
2. An Introduction to Polymer Chemistry, W.R.Moor, London Press, London.
3. Principles of Polymer Systems, Rodri-Guez, McGraw Hill Book Co. New York.
4. Modern Technology of Plastics and Polymer Processing Industries, NIIR Board
5. Sugar: Science and Technology, G. G. Birch and K.. J. Parker, Applied Science Publishers Ltd., 1979.
6. Principles of Sugar Technology, Pieter Honig Vol I, Elsevier Publishing Company, 1953.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-436	Applied Chemistry (Sp. Practical)	2	VIII
Year	Discipline		
4	Chemistry		

### **SYLLABUS OUTLINE:**

#### **Petroleum:**

Determination of diesel index, Aniline point and Pour point of lubricating oil.

#### **Water Pollution:**

Heavy metals in industrial effluents by AAS.

#### **Industrial Analysis:**

Sucrose in sugar cane juice using polarimetry.

Isolation and determination of casein, lactose and fat in milk.

Analysis of effluents from leather tanneries.

Assay of pharmaceutical products like aspirin, paracetamol, and chloramphenicol using spectroscopic techniques.

#### **Spectrometry:**

Simultaneous determination of chromium and manganese in steel.

Determination of pK value of indicators (methyl orange and methyl red).

#### **Polymer:**

Depolymerization of polyethylene terephthalate.

Catalytic degradation of polythene.

### **RECOMMENDED BOOKS:**

1. Metallurgical Analysis, Lord (1893).
2. Applied Chemistry Theory and Practice, O.P. Vermani & A.K. Narula, Wiley Eastern Limited (1989).
3. T. B. of Quantitative Inorganic Analysis, Vogel's Ed-4<sup>th</sup>, Longman Group Limited (1978).
4. Practical Statistics for the Analytical Scientist, A Bench Guide, RSC Publishing, LGC Ltd 2009.
5. Dyes and Dyeing, C. E. Pellew, Abhishek Publishers (1998).
6. Experiments in Physical Chemistry, David P. Shoemaker, McGraw Hill International (1996).

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-418	Biochemistry (Sp. Theory-I)	4	VII
Year	Discipline		
4	Chemistry		

### **SYLLABUS OUTLINE:**

#### **1. Basis of Metabolism:**

Methods of metabolism study. Cell bioenergetics and Role of ATP. Biological oxidation and reduction. Electron Carriers involved in the oxidation of fuel molecules. Oxidative phosphorylation and regulation of ATP production. Inhibitors of electron transport chain. Uncouplers of oxidative phosphorylation.

#### **2. Metabolism of Carbohydrates:**

Digestion, Absorption, and Transport of Carbohydrates. Glycolysis, Citric acid Cycle. HMP pathway. Uronic acid pathway. Gluconeogenesis. Glycogenesis, Glycogenolysis, Photosynthesis and their control. Regulation of carbohydrate metabolism.

#### **3. Metabolism of Lipids:**

Digestion and absorption of Lipids. Detailed Synthesis and Oxidation of fatty acids, Involving Acyl carrier protein and Carnitine carriers. Metabolism of essential fatty acids and their metabolic disorders. Control of fatty acid Metabolism. Ketone Bodies. Phospholipids, steroids and Prostaglandins.

#### **4. Endocrine system:**

Mechanisms of action of hormones. Chemistry, Metabolism and Biological functions of Pancreatic, Pituitary, Gonadal, Adrenal, Thyroid, Parathyroid, Intestinal and Renal hormones. Pheromones. Hormonal control mechanisms.

### **RECOMMENDED BOOKS:**

1. Principles of Biochemistry by Lehninger AL, Nelson DL and Cox MN, 2000 Pub: Worth Publishers
2. Biochemistry by Lubert Stryer(2006) Pub: Freeman and Company
3. Lippincott's Biochemistry by Champe.P C; Harvey. R. A and Ferrier. D. R. 3rd ed., 2004 Pub: J. b. Lippincott Company
4. The Biochemistry of Cell signaling by Ernst J. Helmreich 2001 Pub: Oxford University Press.
5. Harpers Biochemistry, 27th ed. 2006 Pub: McGraw Hill Inc

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-419	Biochemistry (Sp. Theory- II)	4	VII
Year	Discipline		
4	Chemistry		

### SYLLABUS OUTLINE:

#### 1. **Biochemistry of Body Fluids:**

General composition of Blood, blood plasma, blood proteins, formed elements of blood. Biosynthesis and metabolism of Porphyrin and Hemoglobin. Gaseous transport. Coagulation of blood. Normal and Abnormal composition of Urine and its Biochemical effects. Composition of CSF.

#### 2. **Immunochemistry:**

The immune system. Detailed structure, chemistry and synthesis of immunoglobulins. Myeloma and hybridoma immunoglobulins. Immungenes. Complement system. Inflammatory process. Peripheral leucocytes and Macrophages. Abnormalities of the immune system.

#### 3. **Vitamins in health and disease:**

Water soluble and fat soluble vitamins. A discussion of the occurrence, chemistry, metabolism, physiological functions, deficiency symptoms and requirements of vitamins A, B-Complex, C, D, E and K.

#### 4. **Physical Techniques used in Biochemistry:**

General methods for extraction, fractionation and purification of proteins. Principles of chromatography, Gel filtration, ion-exchange chromatography, Affinity chromatography and gas chromatography. HPLC. Polyacrylamide and agarose gel electrophoresis, SDS PAGE, Immunoelectrophoresis. Immunological techniques including RIA, ELISA, Immunohistochemistry, Electrofocussing. Gradient centrifugation. Ultra filtration. Lyophilization. Electron microscopy. Radioisotopes and their applications in Biochemistry. X-ray diffraction

### RECOMMENDED BOOKS:

1. Principles of Biochemistry by Lehninger AL, Nelson DL and Cox MN, 2000 Pub: worth Publishers
2. Biochemistry by Lubert Stryer(2006) Pub: Freeman and Company
3. A biologist's guide to Principles and Techniques of Practical Biochemistry by Bryan L Williams and Keith Wilson Pub: Edward Arnold Ltd.
4. Immunology by J. Kuby 2nd ed. 1996 Pub: W. H. freeman and Co.
5. Harpers Biochemistry, 27th ed. (2006) McGraw Hill Inc.

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-421	Biochemistry (Sp. Practical)	2	VII
Year	Discipline		
4	Chemistry		

### **SYLLABUS OUTLINE:**

1. **Enzymes:**

Isolation of enzyme from different sources, study of different factors like temperature. pH, Concentration of substrate on the properties of Alkaline Phosphatase and LDH. Determination of the kinetic parameters of these enzymes and their mode of inhibition using UV/Visible Spectrophotometer.

2. **Vitamins:**

Estimation of Vitamin A, B1, B2, C and D in food materials by chemical methods and HPLC.

3. **Urine Analysis:**

Analysis of organic constituents in normal and abnormal human urine.

4. **Blood Analysis:**

Analysis of blood constituent like, Calcium, Phosphate, Sugar, Urea, creatine, Biliuribin, Cholesterol, Triglycerides by chemical methods, flame photometry and atomic absorption spectroscopy.

### **RECOMMENDED BOOKS:**

1. Modern Experimental Biochemistry by R. F. Boyer 3rd ed, 2000, Pub: Pearson Education Inc.
2. Practical clinical Biochemistry by Varley. Pub: CBS publisher
3. An Introduction to Practical Biochemistry By D. T. Plummer 3rd ed. (1987)  
Pub: McGraw Hill

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-437	Biochemistry (Sp. Theory-I)	4	VIII
Year	Discipline		
4	Chemistry		

### **SYLLABUS OUTLINE:**

#### **1. Metabolism of Amino acids and Proteins:**

Digestion and Absorption of Proteins. Biosynthesis of essential amino acids. Degradation of amino acids, Urea Cycle Decarboxylation, transamination and deamination reactions of amino acids and their importance. Creatine and creatinine synthesis and secretion.

#### **2. Metabolism of Nucleic Acids:**

Biosynthesis and Catabolism of Purines and Pyrimidines. Biosynthesis of nucleotides. Disorders linked to serum urate levels. Synthesis of RNA and splicing.

#### **3. Industrial Biochemistry:**

Use of Prokaryotes and Fungi in industry. Industrial production of Ethyl Alcohol, Vinegar, Lactic acid and monosodium Glutamate (MSG) using microorganisms. Bacterial application in food industry i.e. use of bacteria in processing and preserving of milk and meat. Bacterial use in other industries as washing powders,

#### **4. Minerals:**

Discussion of micro, macro and alien metals; their occurrence, physiological functions, deficiency symptoms, daily requirement and their metabolism.

### **RECOMMENDED BOOKS:**

1. Principles of Biochemistry by Lehninger AL, Nelson DL and Cox MN, 2000 Pub: worth Publishers
2. Biochemistry by Lubert Stryer(2006) Pub: Freeman and Company
3. Principles of cell and Molecular Biology by Kleinsmith.L. J. and Kish. V. M. 2nd ed. 1995, Pub: Harper Collins
4. Harpers Biochemistry, 27th ed. (2006) McGraw Hill Inc

## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
CHEM-438	Biochemistry (Sp. Theory-II)	4	VIII
Year	Discipline		
4	Chemistry		

### **SYLLABUS OUTLINE:**

#### **1. Physiological Chemistry:**

Structure and detoxification function of liver. Structure of Kidney with special reference of excretion and detoxification function. Muscular contraction and relaxation. Nerve conduction and action potential. Ionization of water, weak Acids and weak Bases. Buffering against pH changes in Biological systems. Water metabolism and acid base balance.

#### **2. Molecular Biology:**

DNA as a genetic material. Eukaryotic chromosomes. Genes and mutational units. Replication, transcription and translation. Protein Synthesis and Genetic code. DNA repair and recombination. Restriction enzymes. Regulation of gene expression in Prokaryotes & Eukaryotes and Operon model. Constitutive, repressed and induced enzymes. Plasmids, bacteriophages, cosmids. Methods of recombinant DNA, Viruses. RNA Processing.

#### **3. Drug Metabolism and chemotherapy:**

Chemistry, metabolism and mechanism of action of antimalarials, antibacterials, antivirals and antifungal drugs. Drug resistance, Biochemical transformation of drugs. Anticancer drugs.

#### **4. Microbial Biochemistry:**

Microorganisms and their gross Classification, Bacterial growth and cultivation techniques. Identification of Microorganisms, Factors for the growth of microbes. Culture medias and their composition. Methods of Growth measurement, Growth under extreme environments. Mutation and protoplast fusion in cultures and its benefits. Gene transfer: transformation, transduction and conjugation. Bacteriophages.

### **RECOMMENDED BOOKS:**

1. Principles of Biochemistry by Lehninger AL, Nelson DL and Cox MN, 2000 Pub: worth Publishers
2. Molecular Biology of the cell by Albert. B and Bray. D 3rd ed. 1994 Pub: Garland Publishing Inc.
3. Principles of cell and Molecular Biology by Kleinsmith.L. J. and Kish. V. M. 2nd ed. 1995, Pub: Harper Collins
4. Harpers Biochemistry, 27th ed. (2006) McGraw Hill Inc
5. Human Physiology- From Cells to Systems by Lauralee Sherwood 4th ed. 2004 Pub: Cole Publishing Co.
6. Fundamentals of Microbiology by E. Alcamo 1994 Pub: Benjamin cummings Co.



## BS (4 Years) for Affiliated Colleges



Code	Subject Title	Cr. Hrs	Semester
<b>CHEM-439</b>	<b>Biochemistry (Sp. Practical)</b>	<b>2</b>	<b>VIII</b>
Year	Discipline		
<b>4</b>	<b>Chemistry</b>		

### **SYLLABUS OUTLINE:**

#### **1. Urine Analysis:**

Analysis of Inorganic constituents in normal and abnormal urine by atomic absorption spectrometry, flame photometry and titration method.

#### **2. Blood Analysis:**

Analysis of inorganic constituents of blood like , sodium, potassium etc by flame photometry. Estimation of organic constituents like Uric acid, serum proteins, haemoglobin etc by chemical methods.

Estimation of Clinically important enzymes like alkaline phosphatase, acid phosphatase, SGPT, SGOT, creatine kinase, etc using their specific assay methods.

##### **(a) Cell structure:**

Study of cell structure by light microscope. Growth of Bacteria and its growth curve.

##### **(b) Separation techniques:**

Gel filtration of proteins, Separation of Blood proteins by Polyacrylamide gel electrophoresis.

### **RECOMMENDED BOOKS:**

1. Modern Experimental Biochemistry by R. F. Boyer 3rd ed, 2000, Pub: Pearson Education Inc.
2. Practical clinical Biochemistry by Varley. Pub: CBS publisher
3. An Introduction to Practical Biochemistry By D. T. Plummer 3rd ed. (1987) Pub: McGraw Hill
4. Fundamentals of Microbiology. By E. Aicamo 1994 Publisher; Benjamin- Cummings Publishing Co.