

<b>Course Title</b>	<b>THERMAL PHYSICS</b>
<b>Course Code</b>	<b>MPHY-213</b>
<b>Credit Hours</b>	<b>CH 3</b>
<b>Pre- requisites</b>	<b>FSc / A-Level (Physics) or equivalent</b>
<b>Learning outcomes</b>	The objective of this course is to develop an understanding about the laws and methods of thermodynamics and enable the student to apply their knowledge to practical systems.
<b>Contents</b>	<p><b>Temperature and heat:</b> Temperature and thermal equilibrium, thermometers, kelvin scales, thermal expansion, Ideal gas, quantity of heat, calorimetry, phase changes, heat transfer.</p> <p><b>Thermal properties:</b> Equations of states, van der Waals equation, molecular properties of matter, molecular view of pressure, mean free path, kinetic model of ideal gas, heat capacities, molecular speeds and energies.</p> <p><b>First law of thermodynamics:</b> Thermodynamics systems, work done, Thermodynamics states, internal energy, Zeroth and First law of Thermodynamics, Thermodynamics processes, internal energy and heat capacities of an ideal gas, Adiabatic processes,</p> <p><b>Second Law of thermodynamics:</b> Heat engines, combustion engines, refrigerator, Second law of thermodynamics, Perpetual motion, Carnot Cycle, Carnot engine, Entropy, Microscopic interpretation of entropy, Efficiencies of real engines, thermoelectricity, Seebeck effect, Peltier effect, thermocouple.</p> <p><b>Applications of thermodynamics:</b> Thermodynamics functions and equations, TdS equations, Joule-Thomson effect, Stephan law, Adiabatic demagnetization, production and measurements of low temperatures, Third law of thermodynamics, Clausius-Clapeyron equation.</p>
<b>Teaching-learning Strategies</b>	Classroom teaching / Lecturing
<b>Assignments- Types and Number</b>	Problem sheets: 3-4
<b>Assessment and Examinations</b>	<p>Mid-Term Assessment: 35%</p> <p>Formative Assessment: (25%): It includes classroom participation, attendance, assignments and presentations, homework, attitude and behavior, hands-on-activities, short tests, quizzes etc.</p> <p>Final Term Assessment: 40%</p>
<b>Text Books</b>	<ol style="list-style-type: none"> <li>1. Physics (Volume 1 &amp; 2) by R. Resnick, D. Halliday and K. S. Krane (5<sup>th</sup> Edition), Wiley (2002)</li> <li>2. Concepts in Thermal Physics, by S. J. Blundell and K. M. Blundell, Oxford, (2009)</li> <li>3. University Physics with Modern Physics by H. D. Young, R. A. Freedman (14<sup>th</sup> Edition), Addison-Wesley (2015).</li> <li>4. Principle of Modern Thermodynamics by B. N. Roy, Institute of Physics, London (1995)</li> <li>5. Physics for Scientists and Engineers by R. A. Serway and J. W. Jewett (8<sup>th</sup> Edition), Golden Sunburst Series (2010).</li> <li>6. An Introduction to Thermal Physics, D. V. Schroeder, Pearson, (1999).</li> <li>7. Heat and Thermodynamics by M. W. Zemansky (7<sup>th</sup> Edition), McGraw Hill (1999).</li> </ol>