Programme	Biochemistry	Course Code	BC. 202	Credit Hours	2+1			
Course Title	Proteins and Nucleic acids	5						
Course Introduction								
This course provides an in-depth introduction to the fundamental building blocks of life, focusing on amino acids and proteins. Students will explore the structure, classification, and properties of amino acids. Additionally, the course introduces nucleic acids, discussing their types, chemistry, functions, and the structural organization of DNA and RNA, including their roles in genetic information storage and transmission.								
	Learnii	ng Outcomes						
On the completion of the course, the students will be able to: • Describe different levels of protein structure • Identify the different amino-acids and nucleic acids • Isolate and analyze the proteins and nucleic acids								
	Cours	se Contents						
Theory Unit-I Introduction to Amino acids Physical and chemical properties and classification of amino acids 20 standard amino acids and their structures Configuration of amino acids, Uncommon amino acids, Amino acids as acids and bases Titration curves of amino acids Peptides and proteins: Protein structure; Primary and secondary structures Super secondary structures of proteins, Motifs and domains Tertiary and quarternary structure of proteins Fibrous and Globular proteins: α -Keratin, Collagen, Elastic, Silk Fibroin Fibrous proteins; Heme proteins, hemoglobin, Myoglobin Allosteric regulation of Hemoglobin, Hemoglobinopathies Reversible binding of proteins with ligands, Cooperative ligand binding Protein interactions mediated by chemical energy; actin and myosin Protein Folding, Molecular chaperones Protein folding, Molecular chaperones Protein denaturation Phosphodiester linkages in nucleic acids, Nucleic acid structure DNA double helix, Watson and Crick Model RNA; structure, function and composition mRNA and its function Practical Unit Qualitative tests for amino acids: Ninhydrin test and Xanthoproteic test, Pauly's diazo test, Sakaguchi test, Millon's test, Hopkin's cole test, Elrich's test, Sodium Nitroprusside test, Lead Sulphide test, Biuret test Separation of forteins form plant sources and their confirmative tests. Determination by using different methods (Bradford, lowery and biuret methods); Protein estimation by using UV/Visible spectrophotometer								

Textbooks and Reading Materials				
• David L. Nelson, Michael M. Cox W.H. Freeman; (November 21, 2012) Lehninger Principles of				
Biochemistry 6th Ed.				
• Laurence A. Moran, Robert A Horton, Gray Scrimgeour and Marc Perry (2011) Principles of				
Biochemistry 5th Ed.				
• D. J. Voet, G.J. Voet and C. W. Pratt. J. Wiley & Sons (2010) Fundamentals of Biochemistry 4th				
Ed. Joshi A. Rashmi. B. Jain Publishers, 2002 Textbook of Practical Biochemistry.				
• S. K. Sawhney, Randhir Singh 2005. <i>Introductory Practical Biochemistry</i> . 2nd Ed Alpha Science				
International, Ltd.				
• By Robert Murray, David Bender, Kathleen M. Botham, Peter J. Kennelly, Victor Rodwell,				
P.Anthony Weil. McGraw-Hill Medical Harpers Illustrated Biochemistry. (2012). 29 th ed				
Teaching Learning Strategies				
• Lectures				
Assignments and Presentations				
Group discussions				
• Interactive sessions				
Assignments: Types and Number with Calendar				
• Ouiz in the Wools of 5 months				

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- Quiz in 4th Week of 5 marks Assigned in 8th Week of 10 marks Presentations in 12th week of 10 marks

Assessment					
Sr. No.	Elements	Weightage	Details		
1	Midterm Assessment	35%	Written Assessment at the mid-point of the semester.		
2	Formative Assessment	25%	Continuous assessment includes: Classroom participation, assignments, presentations, viva voce, attitude and behavior, hands-on-activities, short tests, projects, practical, reflections, readings, quizzes etc.		
3	Final Assessment	40%	Written Examination at the end of the semester. It is mostly in the form of a test, but owing to the nature of the course the teacher may assess their students based on term paper, research proposal development, field work and report writing etc.		