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## **UNIVERSITY OF THE PUNJAB**

B.S. 4 Years Program / Second Semester - Spring 2022

gram / Second Semester – Spring 2022

Paper: Chemistry-II (Inorganic Chemistry) Course Code: CHEM-103

## THE ANSWERS MUST BE ATTEMPTED ON THE ANSWER SHEET PROVIDED

Q.1. **Answer the following short questions:** (15x2=30)(i) How CFT is superior to VBT? How radiations can be measured? (ii) What are the raw materials of soap industry? (iii) How colour is imparted to Glass during its manufacturing process. (iv) What is soda ash? **(v)** (vi) What is carbon dating? What are the ores of copper from which it is extracted? (vii) (viii) What is the shape of SF<sub>4</sub> according to VSEPR theory? What is the example of linkage isomerism? (ix) **(x)** What are the applications of coordination compounds in industry? What are the biological effects of nuclear radiations? (xi) (xii) What are chelates? (xiii) Explain diagonal relationship of Li and Mg. (xiv) O<sub>2</sub> molecule is paramagnetic in nature. Justify the statement. (xv) What are the main defects of VESPR theory? Q.2. **Answer the following questions:** (3X10=30)Explain the structures of Ni(CO)<sub>4</sub> and [Ni(CN)<sub>4</sub>]<sup>2</sup> complexes on the basis of VBT. (i) theory? (ii) How electronegativity can be measured by different methods?

What is the role of isomerization in coordination chemistry?